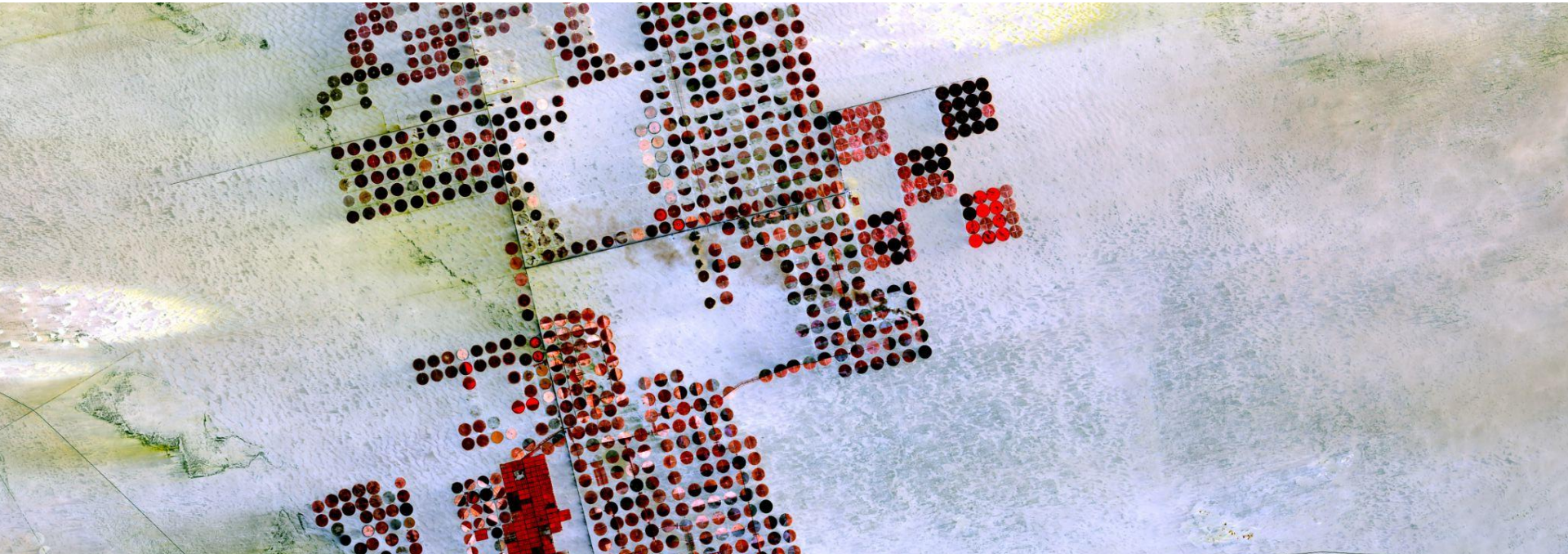


EDS 223: Geospatial Analysis & Remote Sensing

Week 6



Welcome!

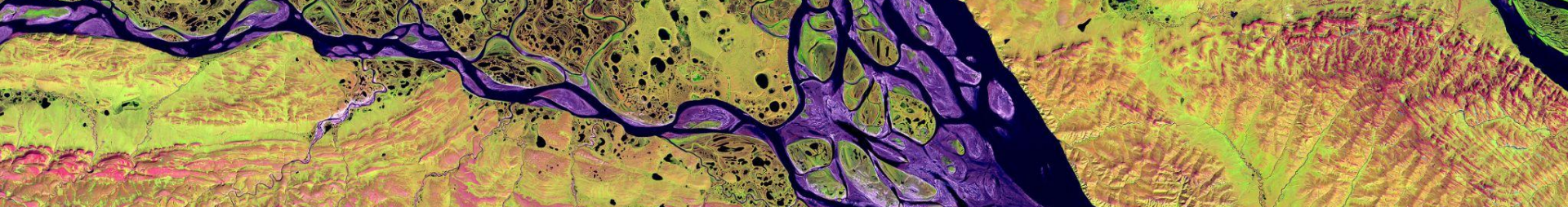
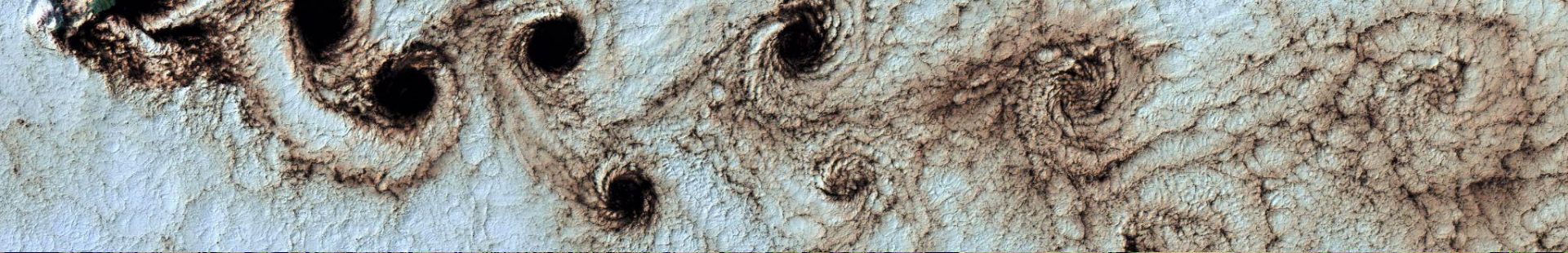
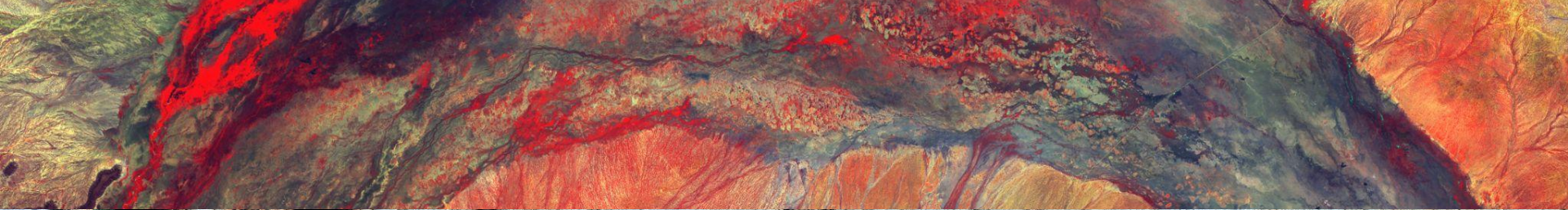
- **Course logistics**
 - Hybrid instruction on November 21
 - non-MEDS students – portfolio proposal

Welcome!

- **Remote sensing basics**
 - Energy transfer
 - Electromagnetic radiation
 - Radiation budget
- **“Self-checks” (aka the opposite of debugging)**
- **Raster geometry operations**

Learning objectives

- **Basic understanding of the physical processes that create remote sensing images**



What is remote sensing?

“the art, science, and technology of obtaining reliable information about physical objects and the environment, through the process of recording, measuring, and interpreting imagery and digital representations of energy patterns derived from non-contact sensor systems.”
(Colwell, 1997)

What is remote sensing?

“the **art, science, and technology** of obtaining reliable information about physical objects and the environment, through the process of recording, measuring, and interpreting imagery and digital representations of energy patterns derived from non-contact sensor systems.”
(Colwell, 1997)

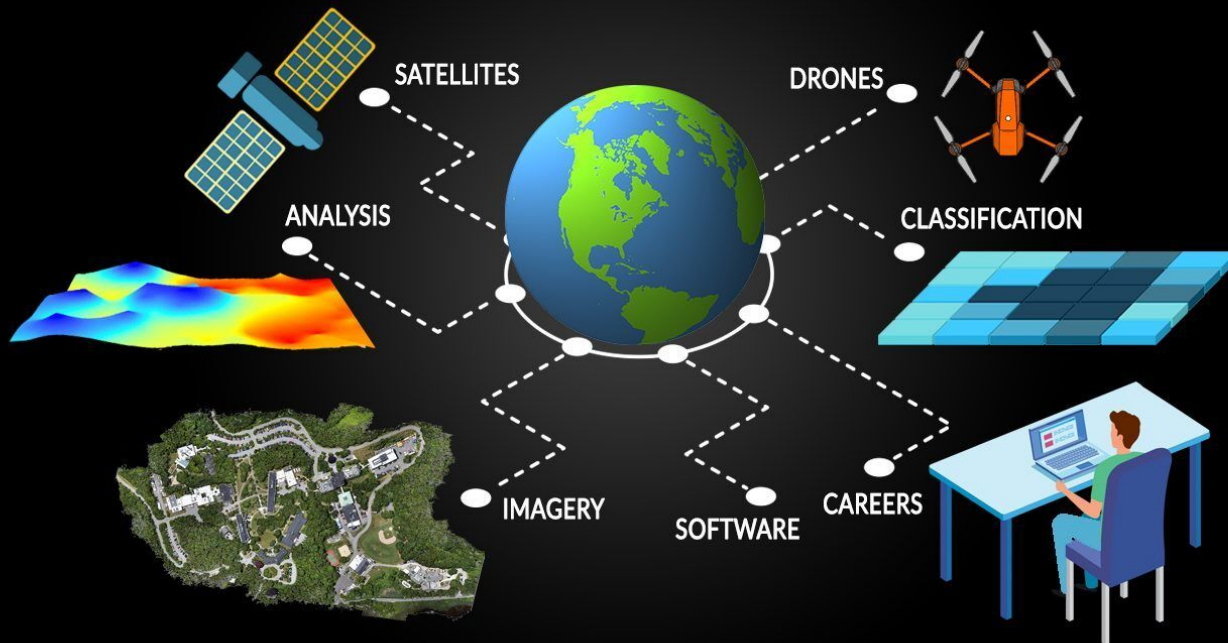
What is remote sensing?

“the **art, science, and technology** of obtaining reliable information about physical objects and the environment, through the process of recording, measuring, and interpreting imagery and digital representations of energy patterns derived from **non-contact sensor systems.**”
(Colwell, 1997)

What is remote sensing?

“the **art, science, and technology** of obtaining reliable information about physical objects and the environment, through the process of recording, measuring, and interpreting imagery and digital representations of **energy** patterns derived from **non-contact sensor systems.**”
(Colwell, 1997)

WHAT IS REMOTE SENSING?

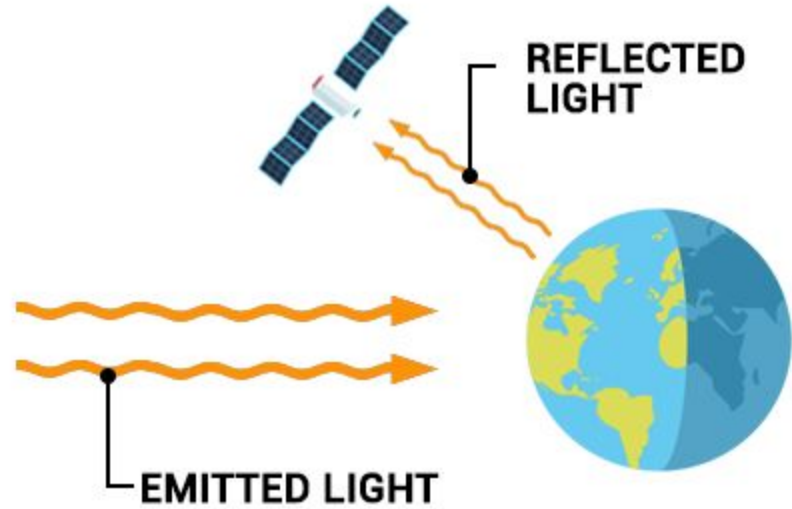


What is remote sensing?

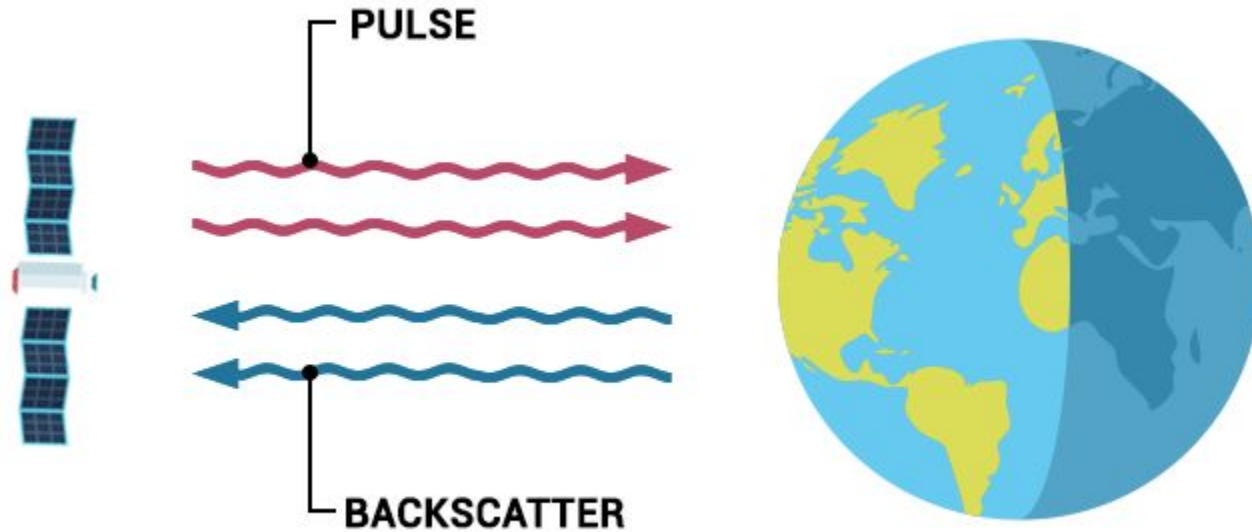
“the **art, science, and technology** of obtaining reliable information about physical objects and the environment, through the process of recording, measuring, and interpreting imagery and digital representations of **energy** patterns derived from **non-contact sensor systems.**”
(Colwell, 1997)

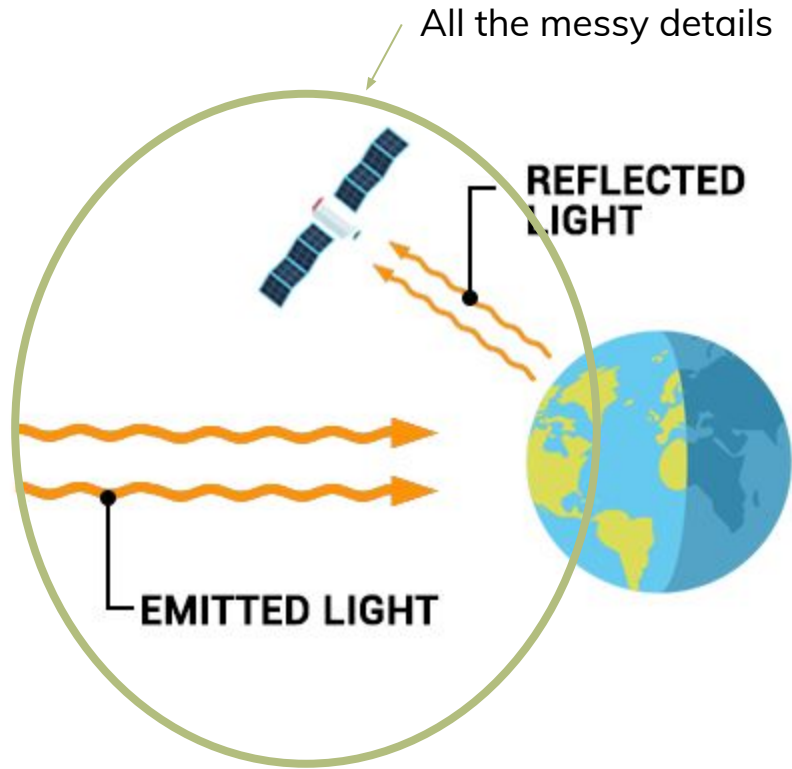
Lots of ways to be a remote sensor, but you are already a remote sensor!

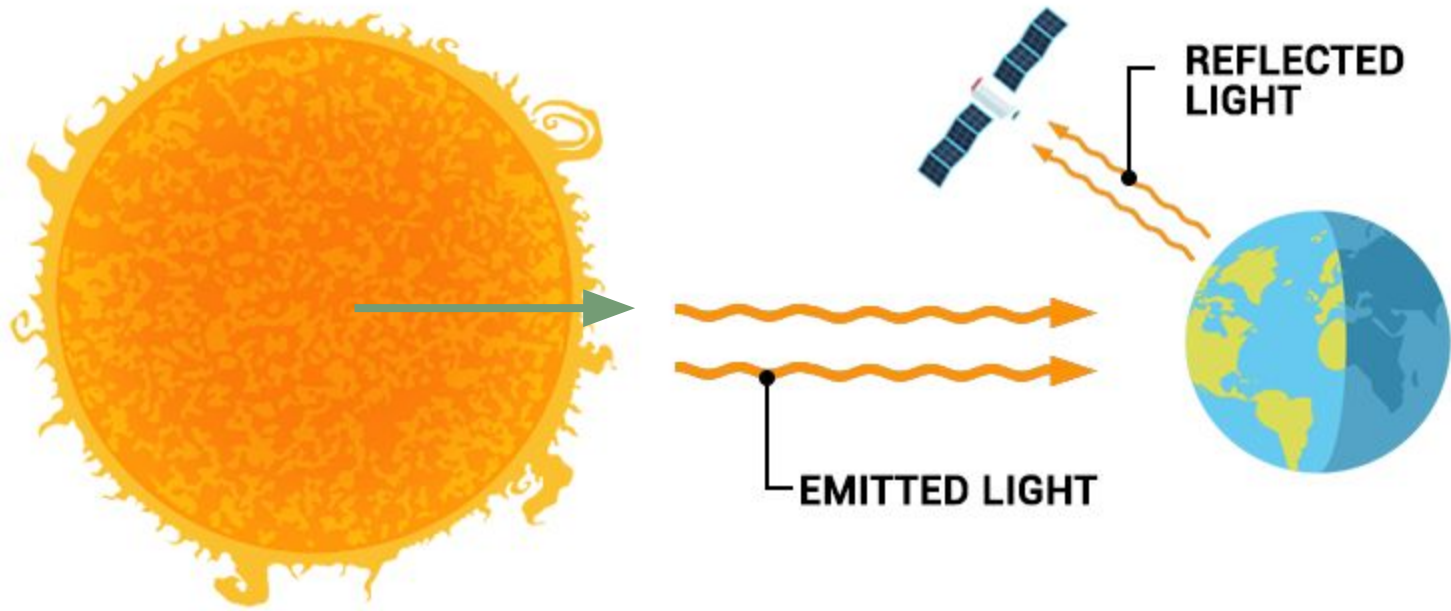
Energy: passive



Energy: active







What is energy?

- The ability to do work (force applied over a distance)
- Lots of different types!
 - kinetic, potential, thermal, gravitational, sound, light, elastic, and electromagnetic energy, etc.
- Any form of energy can be transformed into another form, but the total energy always remains the same (conservation of energy).
 - Energy cannot be destroyed or transferred.

What is energy?

A **newton** is a measure of force...

$$(F = ma)$$

the force needed to accelerate 1
kg of mass as the rate of 1 m/s^2

$$1 \text{ N} = 1 \frac{\text{kg m}}{\text{s}^2}$$

What is energy?

A **newton** is a measure of force...

$$(F = ma)$$

the force needed to accelerate 1 kg of mass as the rate of 1 m/s^2

A **joule** is a measure of energy...

the energy transferred to (or work done on) an object when a force of 1 N acts on an object through a distance of 1 m

$$1 \text{ N} = 1 \frac{\text{kg m}}{\text{s}^2}$$

$$\begin{aligned} 1 \text{ J} &= 1 \text{ N m} \\ &= 1 \frac{\text{kg m}^2}{\text{s}^2} \end{aligned}$$

What is energy?

A **newton** is a measure of force...
($F = ma$)
the force needed to accelerate 1
kg of mass as the rate of 1 m/s^2

$$1 \text{ N} = 1 \frac{\text{kg m}}{\text{s}^2}$$

A **joule** is a measure of energy...
the energy transferred to (or work
done on) an object when a force
of 1 N acts on an object through a
distance of 1 m

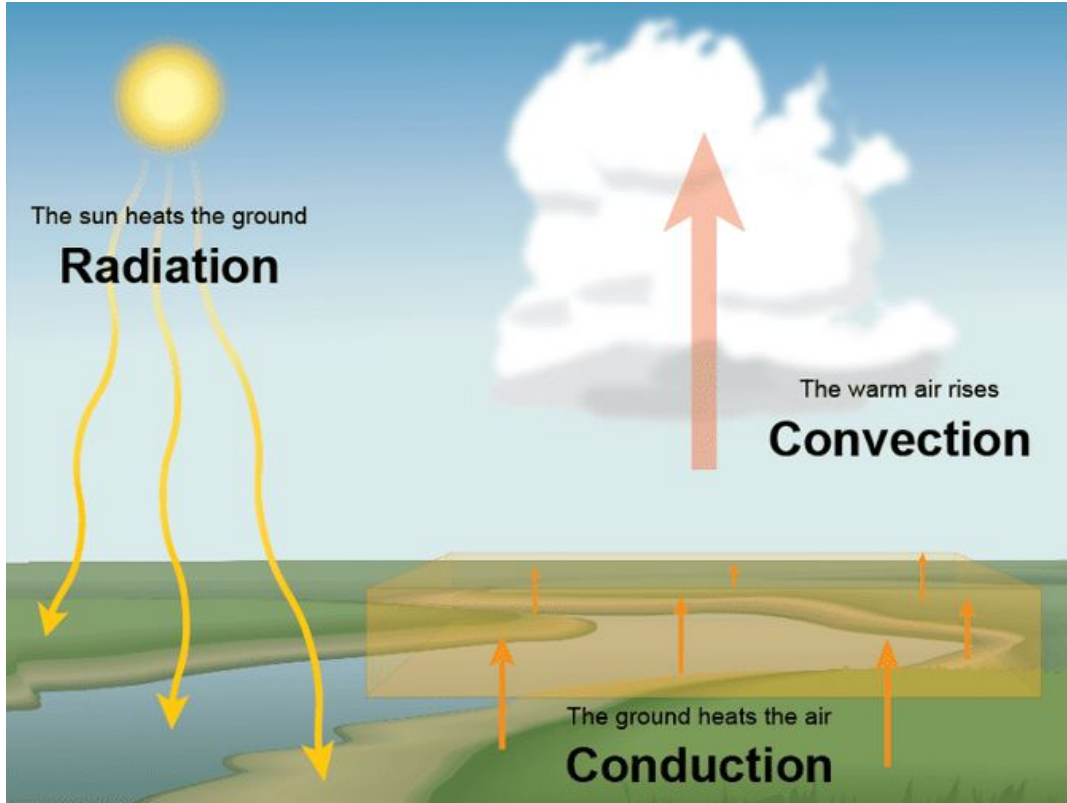
$$\begin{aligned} 1 \text{ J} &= 1 \text{ N m} \\ &= 1 \frac{\text{kg m}^2}{\text{s}^2} \end{aligned}$$

A **watt** is a measure of power...
the rate of energy transfer

$$\begin{aligned} 1 \text{ W} &= 1 \text{ J/s} \\ &= 1 \frac{\text{kg m}^2}{\text{s}^3} \end{aligned}$$

How is energy transferred?

How is energy transferred?



Conduction: A body/object transfers its kinetic energy to another by colliding with it.

Convection: Kinetic energy transferred by physically moving objects.

Radiation: emission of energy as electromagnetic waves or as moving subatomic particles

What is radiation?

Fact: All materials with temperature above absolute zero emit radiation.

What is radiation?

Fact: All materials with temperature above absolute zero emit radiation.

What is **temperature** again?

What is radiation?

Fact: All materials with temperature above absolute zero emit radiation.

What is **temperature** again?

A measure of heat, which reflects the **kinetic energy** of the vibrating and colliding atoms making up a substance.

What is radiation?

Fact: All materials with temperature above absolute zero emit radiation.

What is **temperature** again?

A measure of heat, which reflects the **kinetic energy** of the vibrating and colliding atoms making up a substance.

What is **kinetic energy**?

What is radiation?

Fact: All materials with temperature above absolute zero emit radiation.

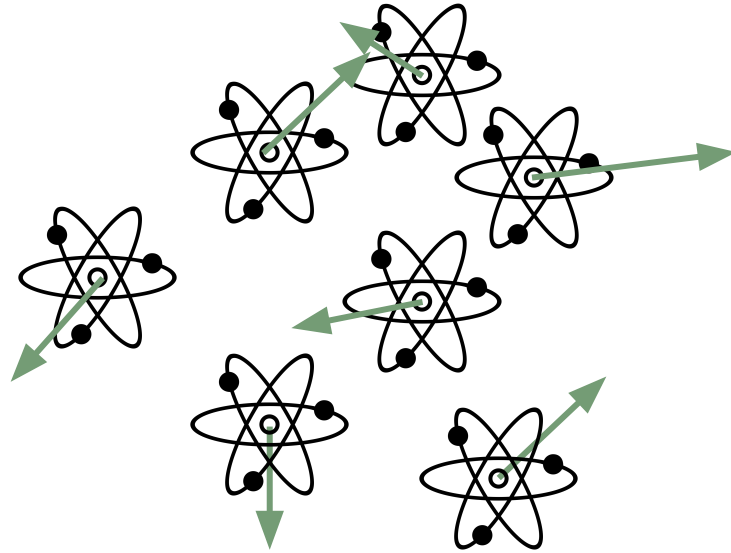
What is **temperature** again?

A measure of heat, which reflects the **kinetic energy** of the vibrating and colliding atoms making up a substance.

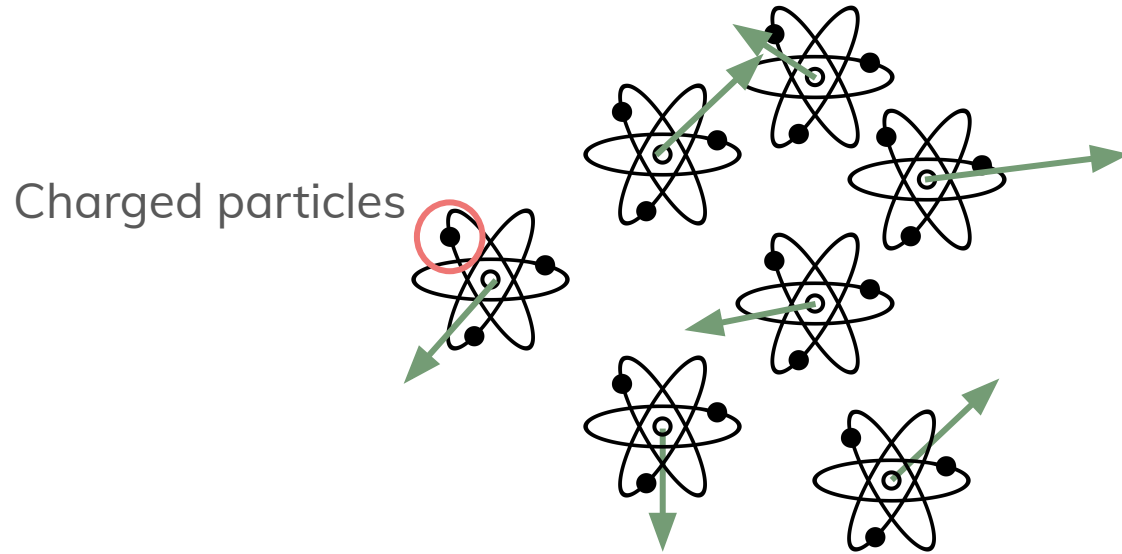
What is **kinetic energy**?

Movement!

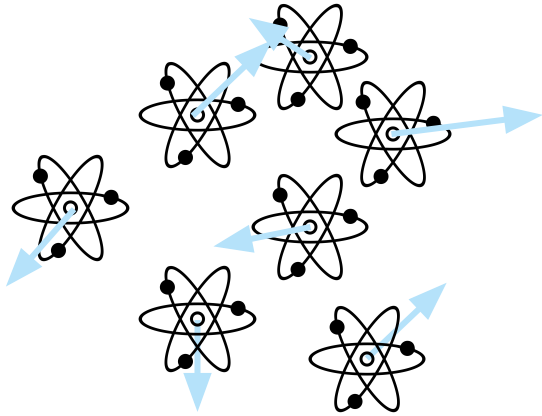
What is radiation?



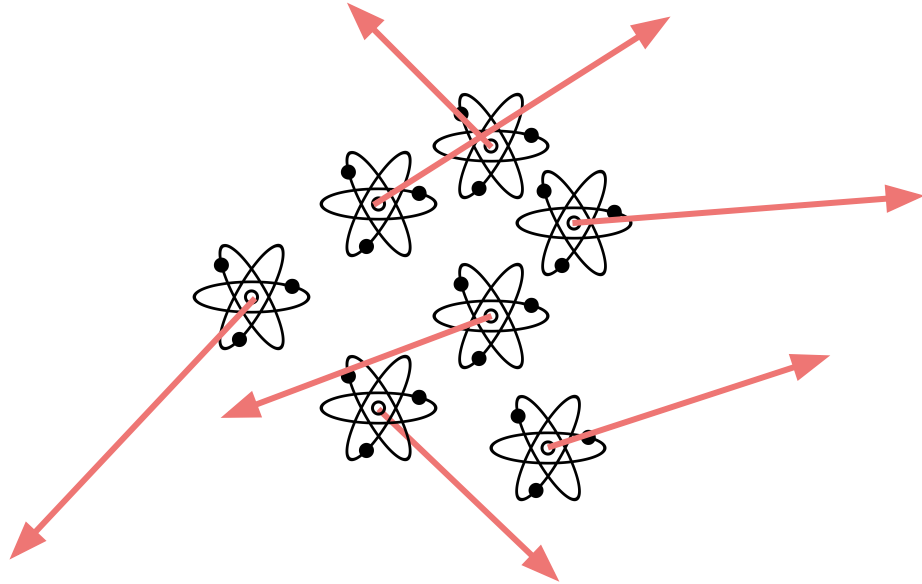
What is radiation?



How is radiation related to temperature?



Low temperature
Less movement
Less kinetic energy



High temperature
More movement
More kinetic energy

How is radiation related to temperature?

As temperature increases, emitted radiation....

- (a) Stays the same
- (b) Increases
- (c) Decreases
- (d) Isn't even paying attention

How is radiation related to temperature?

As temperature increases, emitted radiation....

- (a) Stays the same
- (b) Increases**
- (c) Decreases
- (d) Isn't even paying attention

Total emitted radiation ~ temperature

How is radiation related to temperature?

As temperature increases, emitted radiation....

- (a) Stays the same
- (b) Increases**
- (c) Decreases
- (d) Isn't even paying attention

Total emitted radiation ~ temperature

But, by how much?

How is radiation related to temperature?

- **Blackbody**
 - a theoretical object which radiates energy with perfect efficiency
 - No radiation passes through it and none is reflected
 - Emits all energy
- **Total emitted radiation from a blackbody**

$$M_{\lambda} = \sigma T^4$$

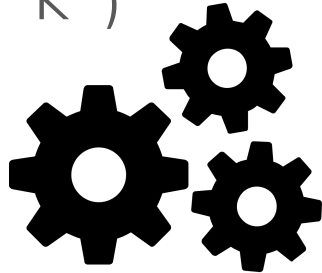
(σ is the Stefan-Boltzmann constant: $5.6697 \times 10^{-8} \text{W m}^{-2}\text{K}^{-4}$)

How is radiation related to temperature?

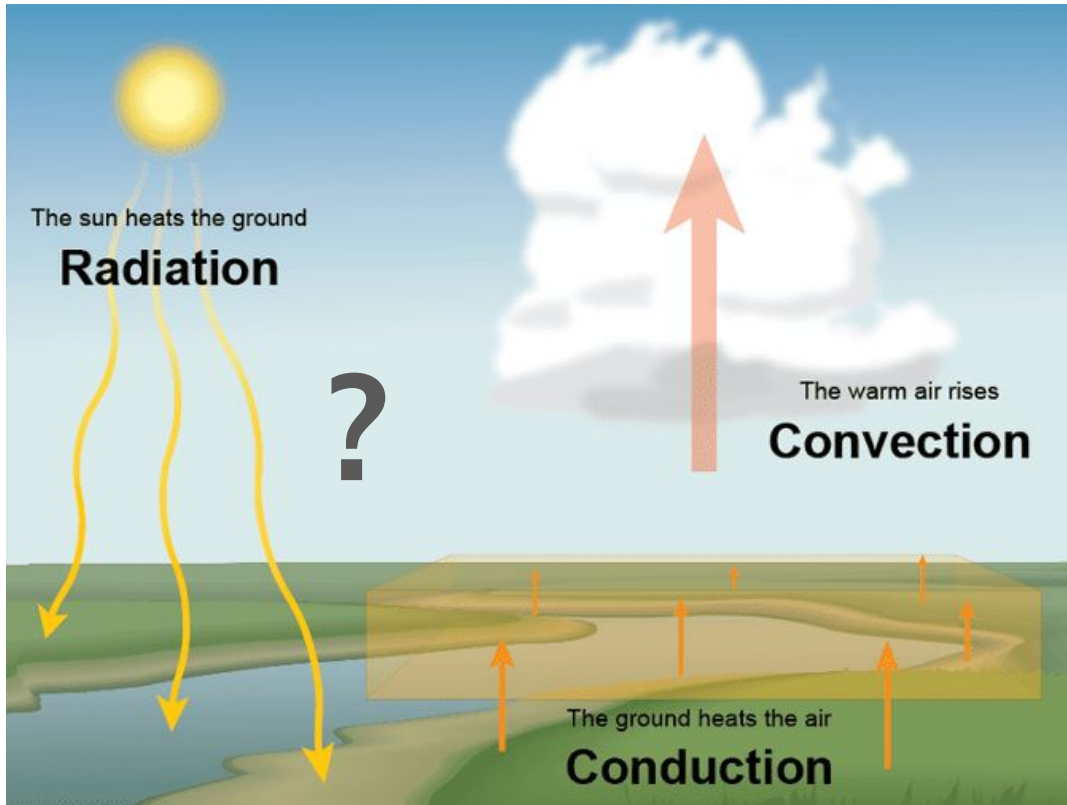
- **Blackbody**
 - a theoretical object which radiates energy with perfect efficiency
 - No radiation passes through it and none is reflected
 - Emits all energy
- **Total emitted radiation from a blackbody**

$$M_{\lambda} = \sigma T^4$$

(σ is the Stefan-Boltzmann constant: $5.6697 \times 10^{-8} \text{W m}^{-2}\text{K}^{-4}$)



How is energy transferred?



Conduction: A body/object transfers its kinetic energy to another by colliding with it.

Convection: Kinetic energy transferred by physically moving objects.

Radiation: emission of energy as electromagnetic waves or as moving subatomic particles

How is radiation transferred?

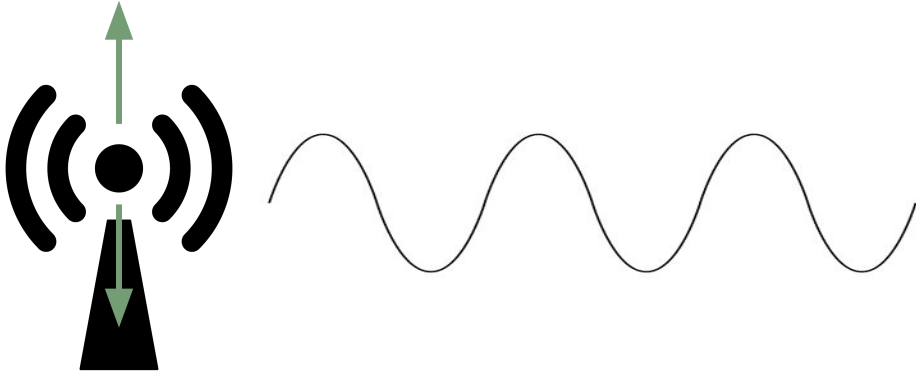
- We know that radiation is produced by the movement of charged particles, but how is transferred??

How is radiation transferred?

- We know that radiation is produced by the movement of charged particles, but how is transferred??
- **We don't really know!**
 - But we've come up with 2 analogies that help us make sense of this:
 - Wave vs. particle model
 - Like all models, neither is perfect but both are useful!

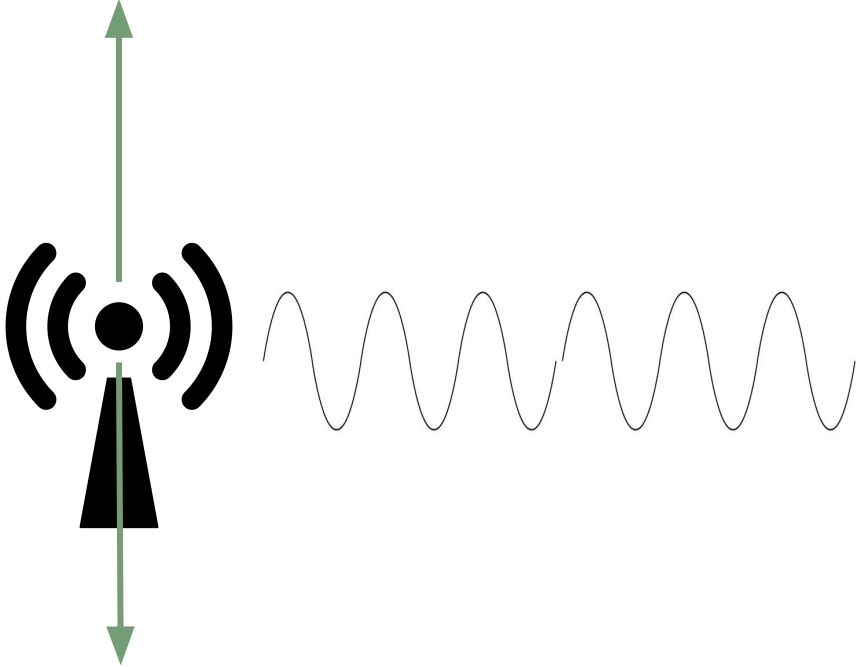
Wave model of radiation

- Generated when a charged particle changes velocity

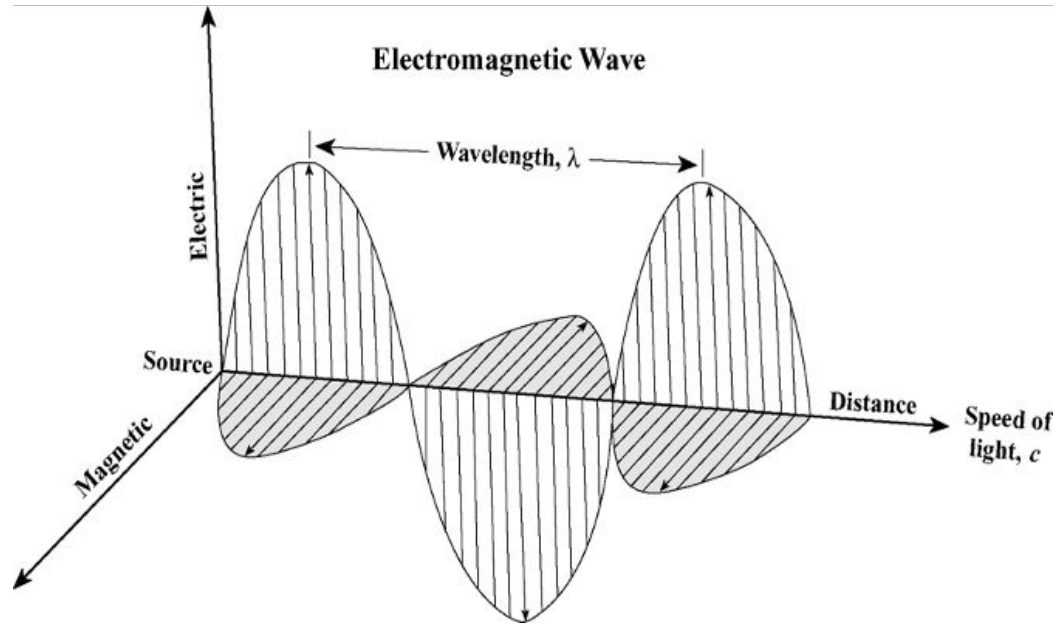


Wave model of radiation

- Generated when a charged particle changes velocity



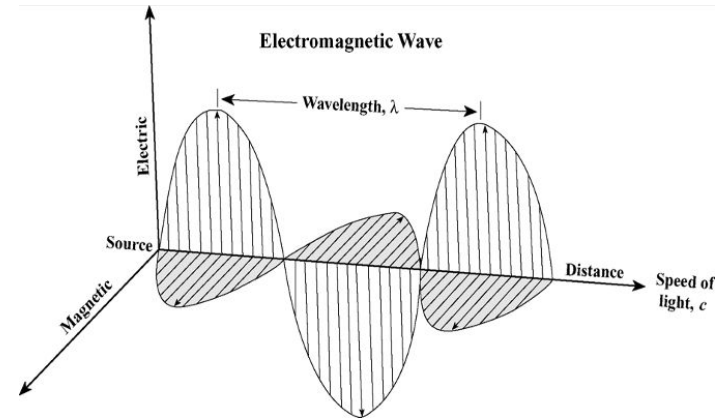
Wave model of electromagnetic radiation



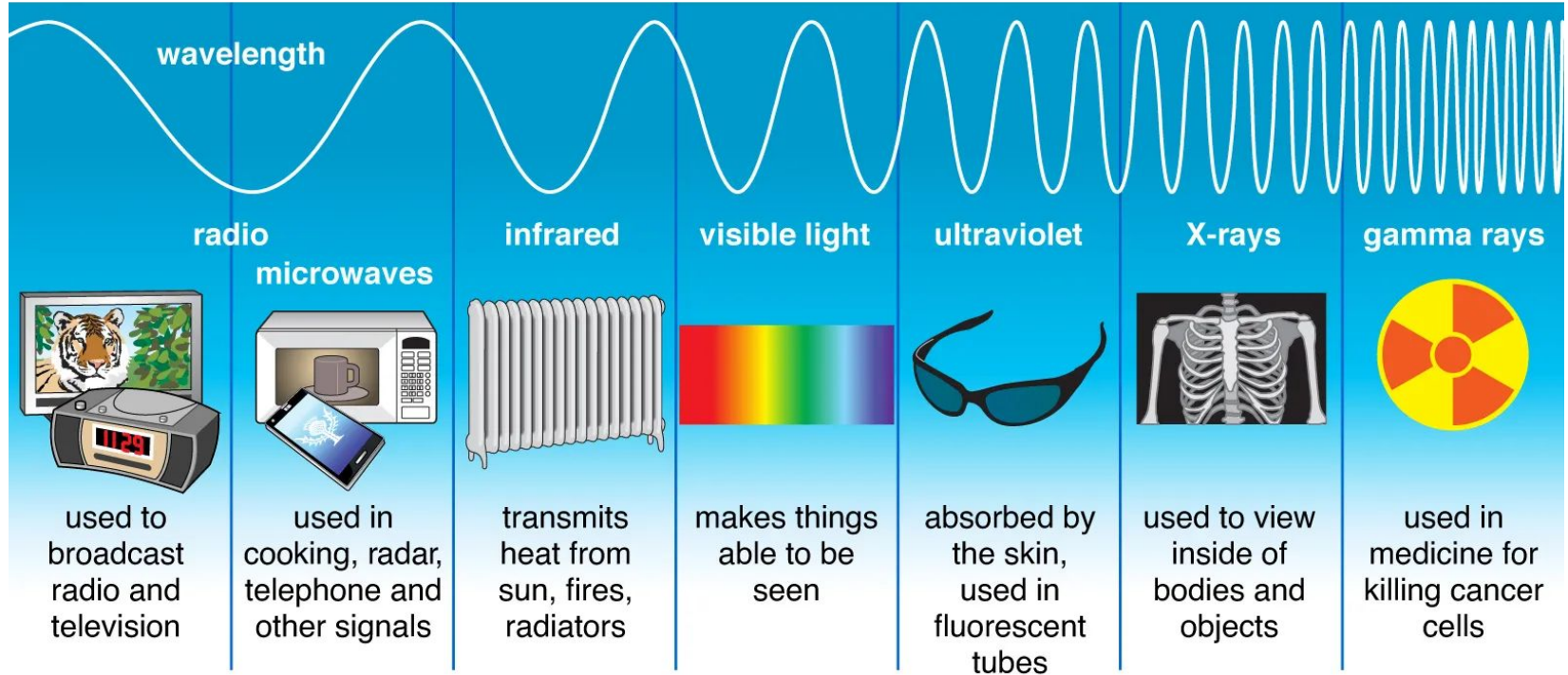
Wave model of electromagnetic radiation

- James Maxwell conceptualized electromagnetic radiation (EMR) as an electromagnetic wave that travels through space at the speed of light
 - c (3×10^8 m/s)
- **Wavelength** (λ) - distance between maximums (or minimums) of a roughly periodic pattern
 - measured in micrometers (μm) or nanometers (nm).
- **Frequency** (ν) - # of wavelengths that pass a point per unit time
 - measured in cycles per second or Hertz (Hz).

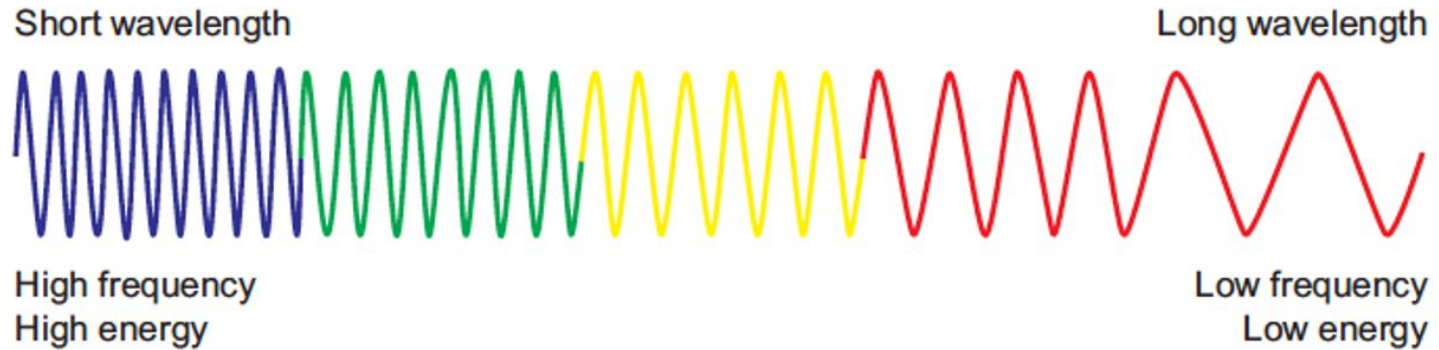
$$c = \lambda \nu$$



Wave model of radiation

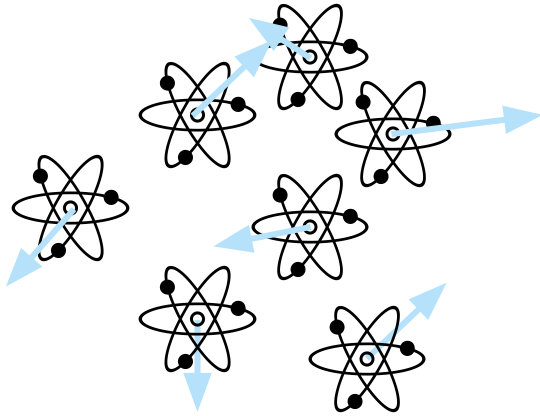


Wave model of electromagnetic radiation

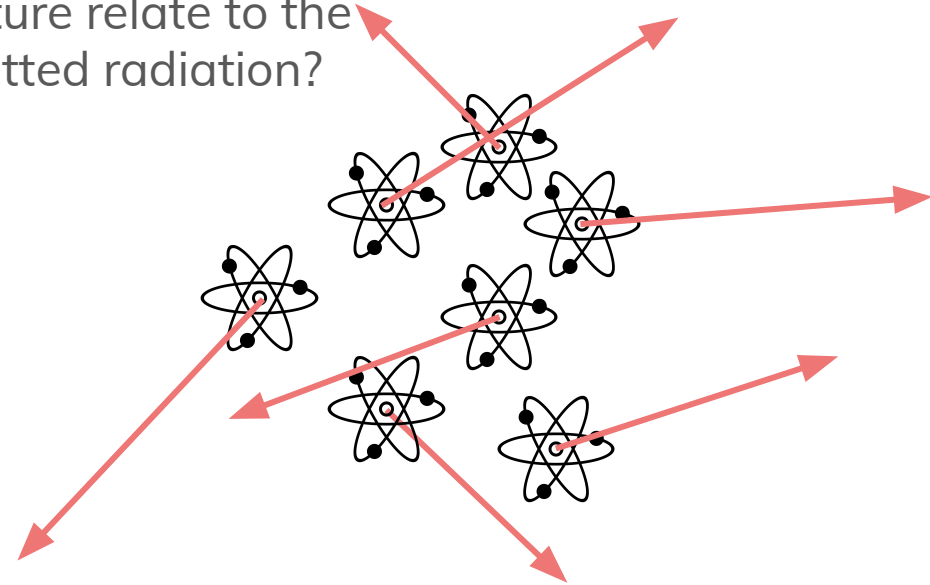


How is radiation related to temperature?

How does temperature relate to the wavelength of emitted radiation?



Low temperature
Less movement
Less kinetic energy



High temperature
More movement
More kinetic energy

How is radiation related to temperature?

As temperature increases, the wavelength of emitted radiation....

- (a) Stays the same
- (b) Increases
- (c) Decreases
- (d) Isn't even paying attention

How is radiation related to temperature?

As temperature increases, the wavelength of emitted radiation....

- (a) Stays the same
- (b) Increases
- (c) Decreases**
- (d) Isn't even paying attention

How is radiation related to temperature?

As temperature increases, the wavelength of emitted radiation....

- (a) Stays the same
- (b) Increases
- (c) Decreases**
- (d) Isn't even paying attention

wavelength \sim 1/temperature

How is radiation related to temperature?

As temperature increases, the wavelength of emitted radiation....

- (a) Stays the same
- (b) Increases
- (c) Decreases**
- (d) Isn't even paying attention

wavelength \sim 1/temperature

But, by how much?

How is radiation related to temperature?

As temperature increases, the wavelength of emitted radiation....

- (a) Stays the same
- (b) Increases
- (c) Decreases**
- (d) Isn't even paying attention

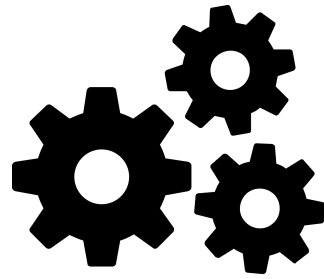
wavelength \sim 1/temperature

But, by how much?

Wien's displacement law:

$$\lambda_{\max} = \frac{k}{T} \quad k = 2898 \mu\text{m K}$$

How is radiation related to temperature?



As temperature increases, the wavelength of emitted radiation....

- (a) Stays the same
- (b) Increases
- (c) **Decreases**
- (d) Isn't even paying attention

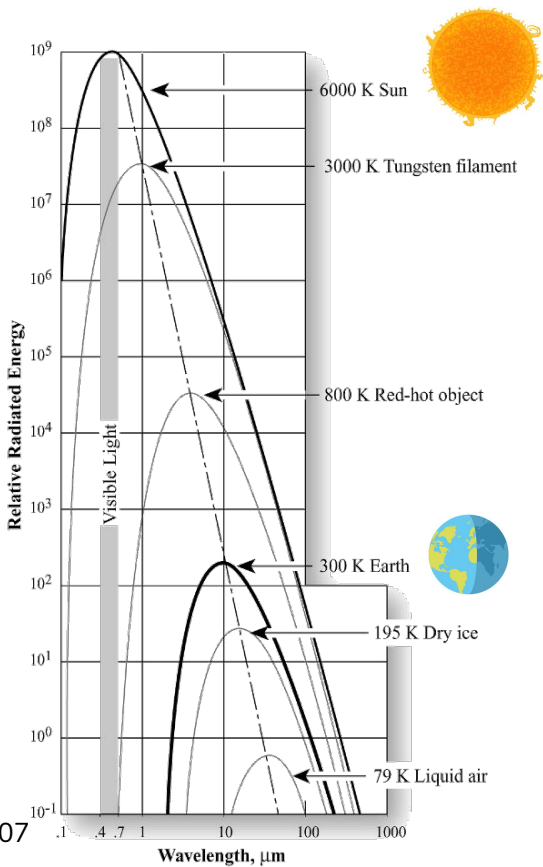
wavelength \sim **1/temperature**

But, by how much?

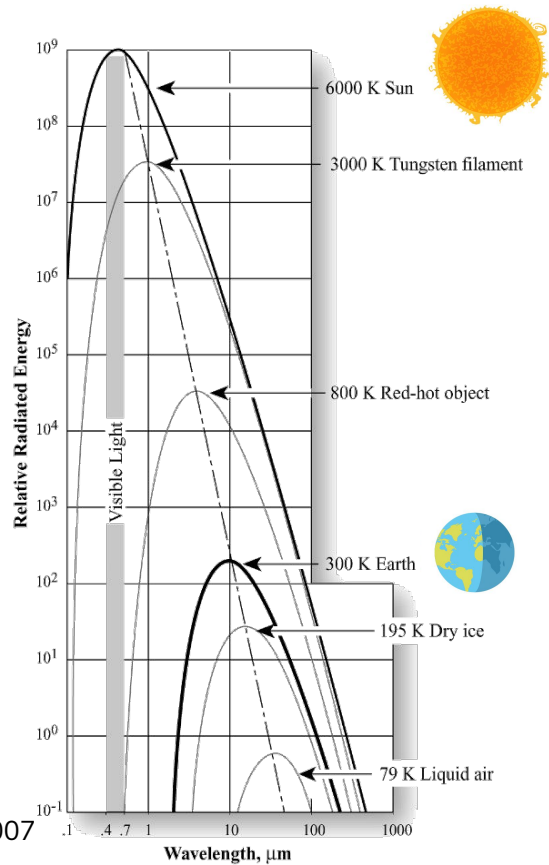
Wien's displacement law:

$$\lambda_{\max} = \frac{k}{T} \quad k = 2898 \mu\text{m K}$$

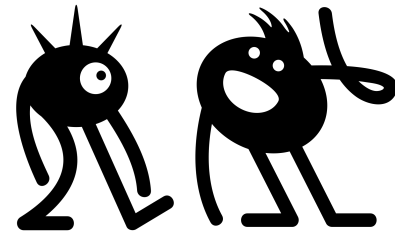
Wave theory: blackbody radiation



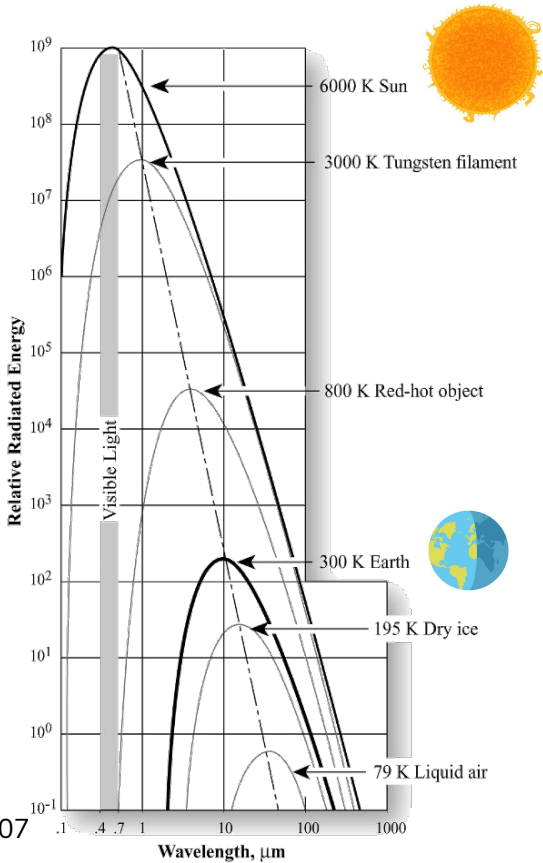
Wave theory: blackbody radiation



Source: Jensen 2007



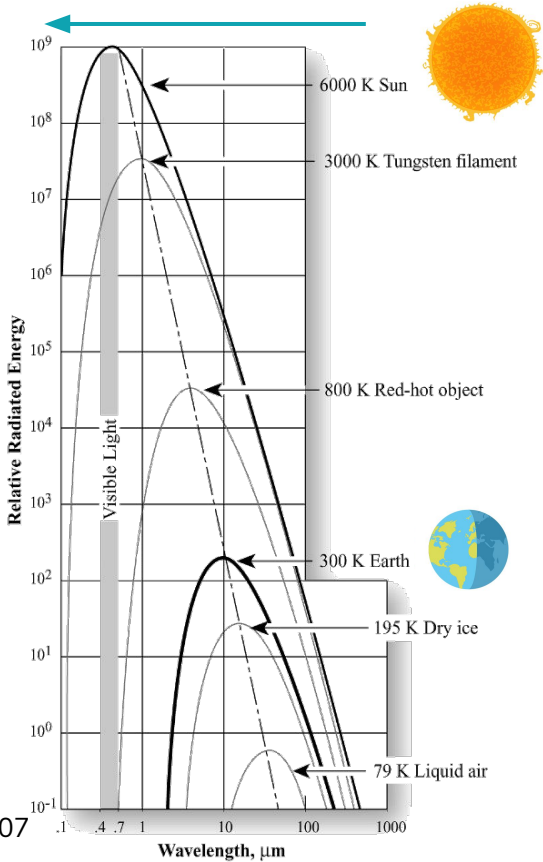
Wave theory: blackbody radiation



- As **temperature** increases, **radiated energy** increases
 - Total energy emitted is equal to the area under the curve

Wave theory: blackbody radiation

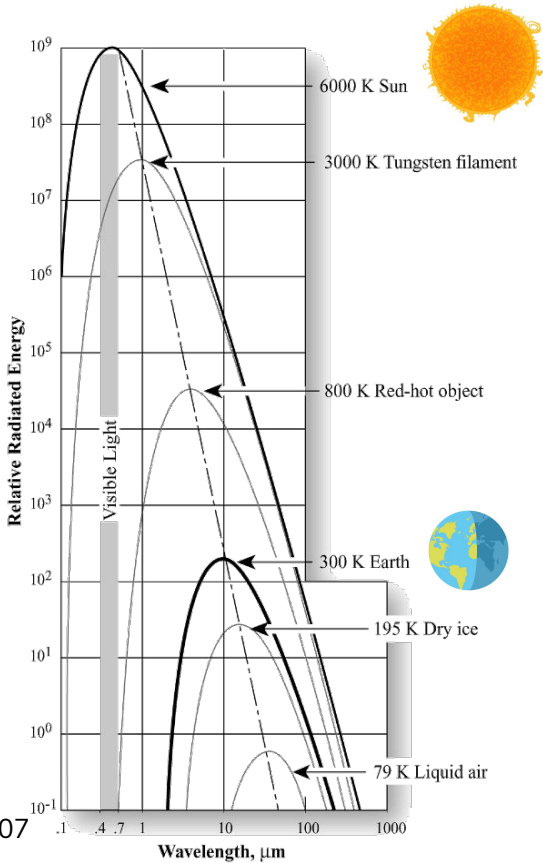
dominant wavelength



- As **temperature** increases, **radiated energy** increases
 - Total energy emitted is equal to the area under the curve
- As **temperature** increases, the **dominant wavelength** of the radiation decreases

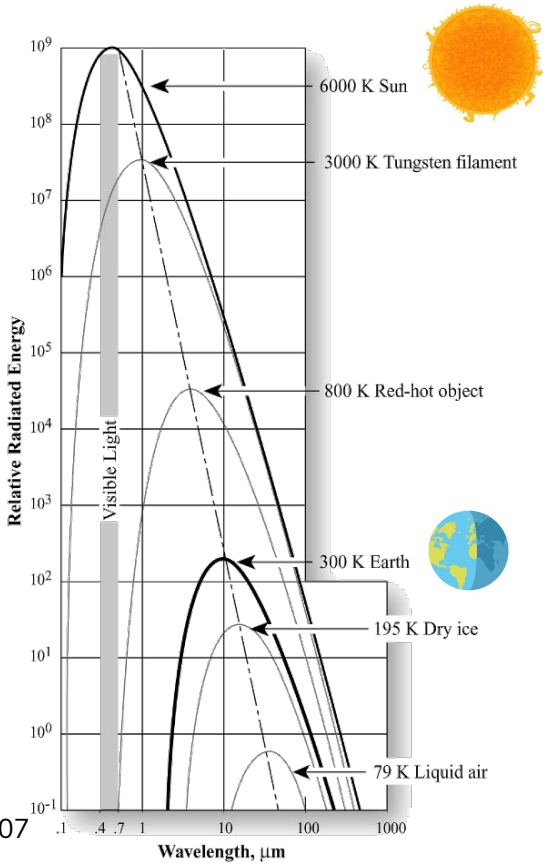
temperature

Wave theory: blackbody radiation



- As **temperature** increases, **radiated energy** increases
 - Total energy emitted is equal to the area under the curve
 - Stefan-Boltzmann's Law
- As **temperature** increases, the **dominant wavelength** of the radiation decreases
 - Wien's Displacement Law

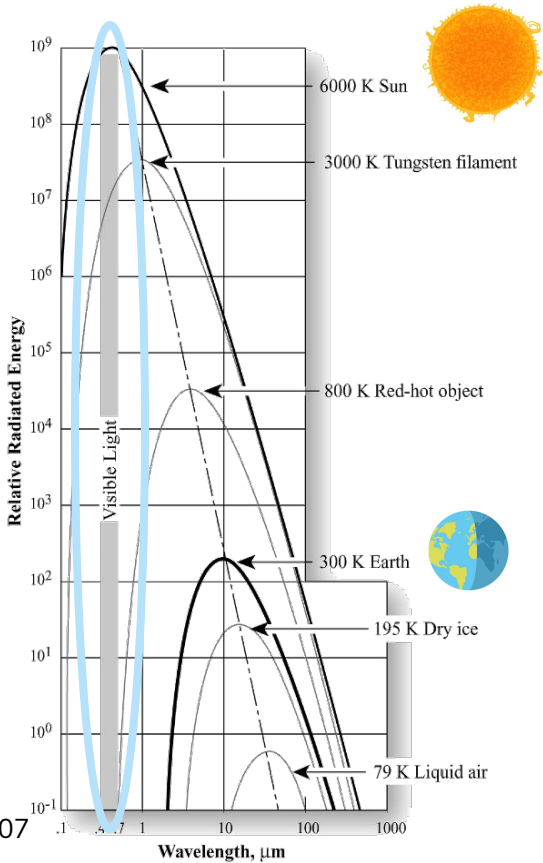
Wave theory: blackbody radiation



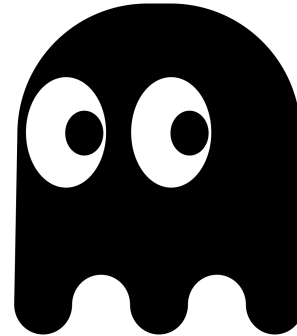
- As **temperature** increases, **radiated energy** increases
 - Total energy emitted is equal to the area under the curve
 - Stefan-Boltzmann's Law
- As **temperature** increases, the **dominant wavelength** of the radiation decreases
 - Wien's Displacement Law



Wave theory: blackbody radiation



- The Sun produces 41% of its energy between 0.4 and 0.7 μm (blue to red light)
- Human eyes are only sensitive to light between 0.4 and 0.7 μm



Wave theory: Stefan-Boltzmann

- **Very few objects on Earth are blackbodies**
 - Instead, we need to know an object's ability to radiate energy
 - emissivity (ϵ): scale of 0 to 1, where a blackbody's emissivity is 1

$$M_{\lambda} = \sigma T^4 \epsilon$$

(σ is the Stefan-Boltzmann constant: $5.6697 \times 10^{-8} \text{W m}^{-2}\text{K}^{-4}$)

POINT: the total amount of radiation energy emitted by an object is proportional to its temperature, and modified by its emissivity

BREAK

The issues...

- **Many possible solutions to one problem**
 - Why use one approach over another?
- **Some solutions will get you to the right answer but need more work**
 - Take the case of “st_join” from assignment 2
- **Unsatisfying answers from the instructor and TA**
 - “Can I use X for this question?” “Sure!” “But the answers wrong...”

The issues...

- **Many possible solutions to one problem**
 - Why use one approach over another?
- **Some solutions will get you to the right answer but need more work**
 - Take the case of “st_join” from assignment 2
- **Unsatisfying answers from the instructor and TA**
 - “Can I use X for this question?” “Sure!” “But the answers wrong...”

What’s our goal with all of this again?

The issues...

- **Many possible solutions to one problem**
 - Why use one approach over another?
- **Some solutions will get you to the right answer but need more work**
 - Take the case of “st_join” from assignment 2
- **Unsatisfying answers from the instructor and TA**
 - “Can I use X for this question?” “Sure!” “But the answers wrong...”

What’s our goal with all of this again?

Building code heroes!

Which means thinking critically about problems.

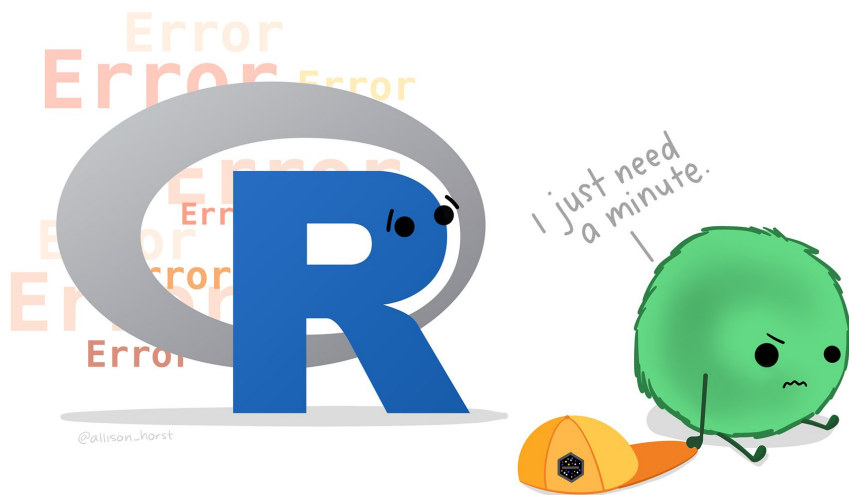
It’s about the process, not the end point!



What's our solution?

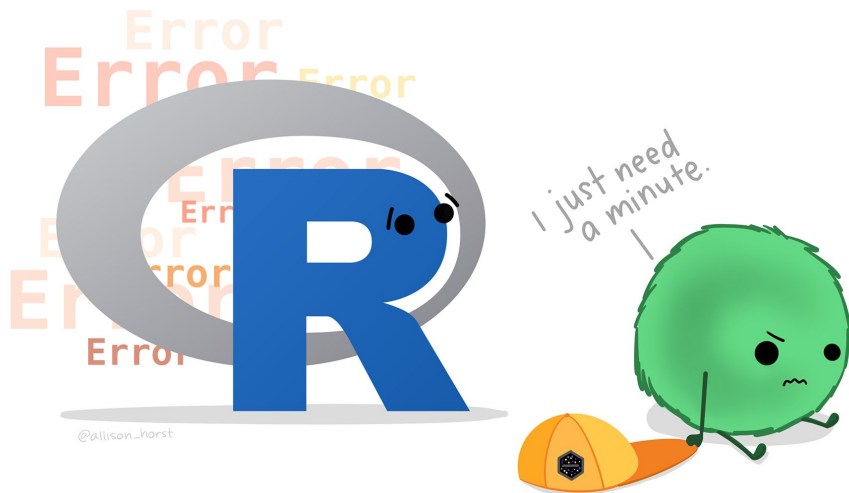
- **Adding critical thinking directly into the rubric for assignments!**
- **Aiming to help with...**
 - Building a better understanding of what your code does and why
 - Reducing frustrating situations with the instructional team :)

Building self-checks into your code



Bugs are scary,
but code that runs but doesn't do
what you think it does is even
SCARIER....

Building self-checks into your code

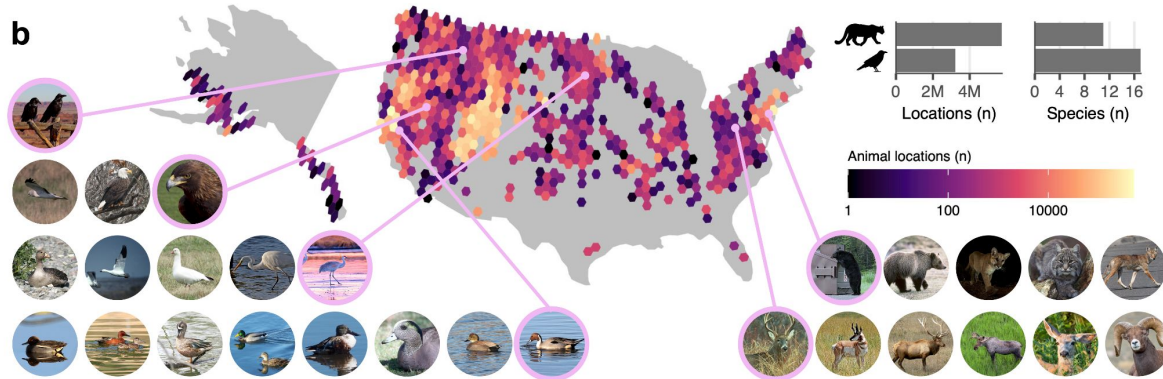
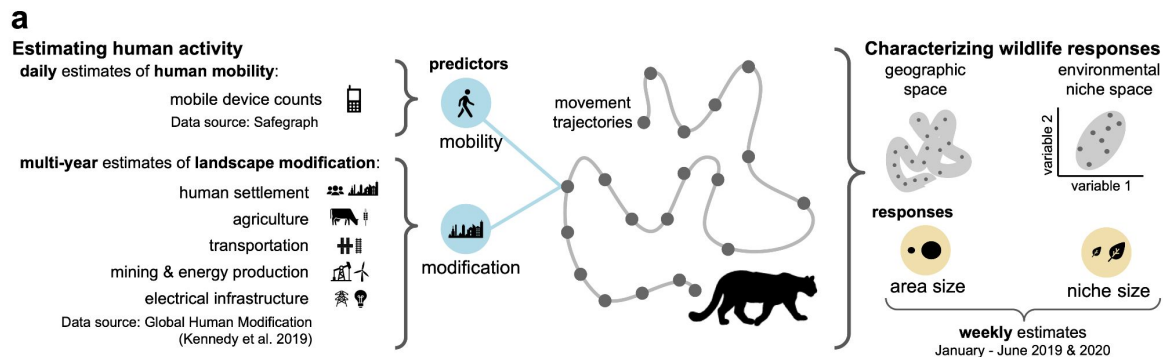


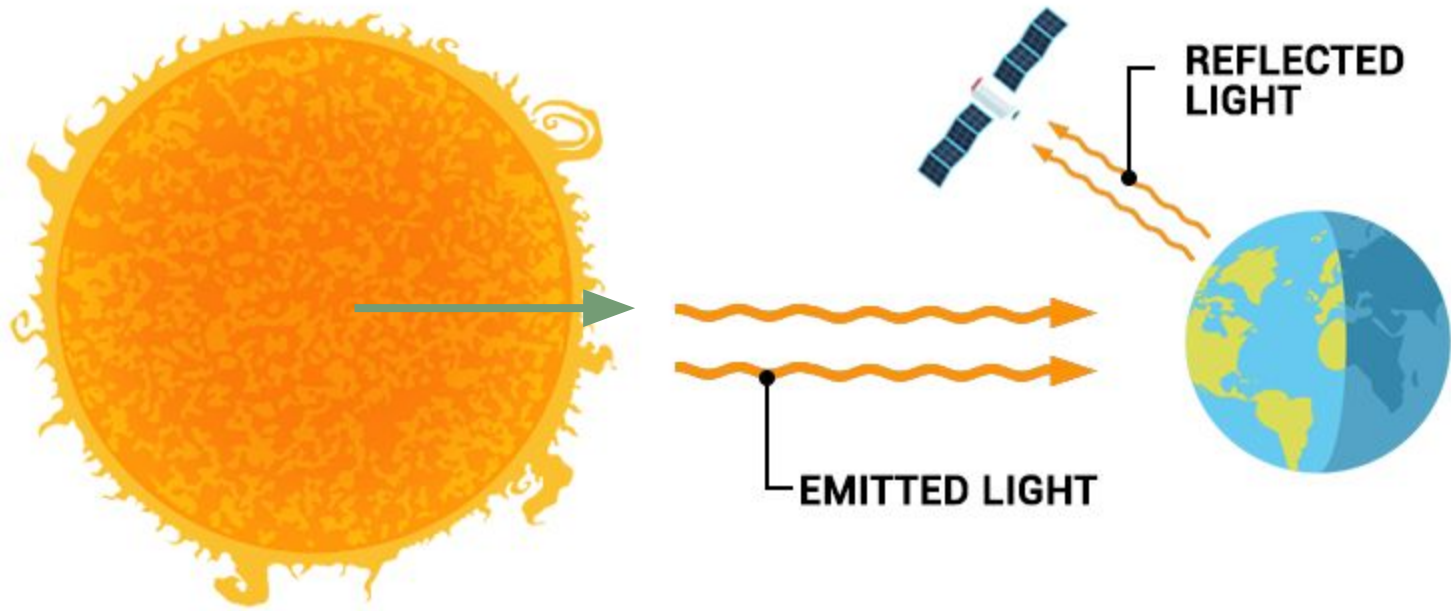
Bugs are scary,
but code that runs but doesn't do
what you think it does is even
SCARIER....

Because... figuring it out takes a
lot of dissecting!

Self-check tutorial

Human mobility drives animals' interactions with space and environment





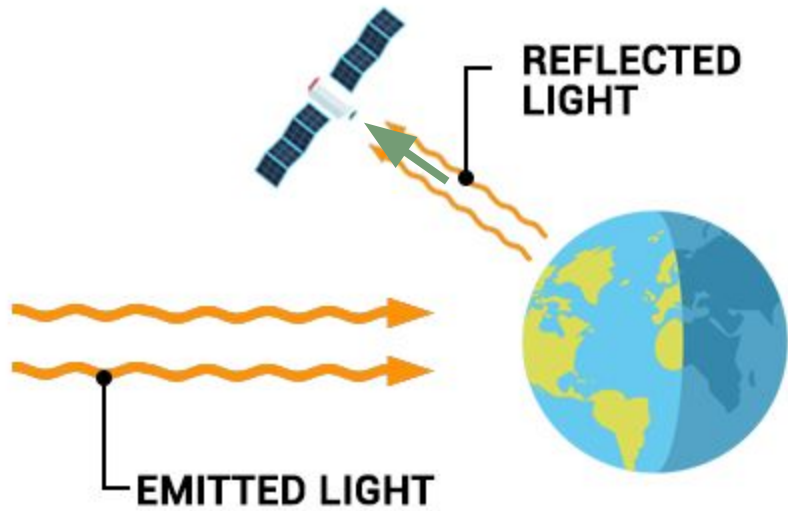
Particle model

- **Niels Bohr and Max Planck proposed the quantum theory of electromagnetic radiation:**
 - Energy is transferred in discrete packets called quanta or photons
- **We can relate the wave and particle models**
 - The energy of a quantum (Q, measured in joules) is related to the frequency of the radiation (ν)

$$Q = h\nu$$

(h is the Planck constant: $6.63 \times 10^{-34} \text{ J s}^{-1}$)

All remote sensing instruments, including cameras, measure the energy of photons, not of waves



What is energy?

A **newton** is a measure of force...
($F = ma$)
the force needed to accelerate 1
kg of mass as the rate of 1 m/s^2

$$1 \text{ N} = 1 \frac{\text{kg m}}{\text{s}^2}$$

A **joule** is a measure of energy...
the energy transferred to (or work
done on) an object when a force
of 1 N acts on an object through a
distance of 1 m

$$\begin{aligned} 1 \text{ J} &= 1 \text{ N m} \\ &= 1 \frac{\text{kg m}^2}{\text{s}^2} \end{aligned}$$

A **watt** is a measure of power...
the rate of energy transfer

$$\begin{aligned} 1 \text{ W} &= 1 \text{ J/s} \\ &= 1 \frac{\text{kg m}^2}{\text{s}^3} \end{aligned}$$

Measuring energy

	Term	Symbol	Units
	Radiant energy	Q	J (Joules)
	Radiant flux	ϕ	W (Watts, J/s)
Radiant flux density	Irradiance	E	W/m ²
	Radiant exitance	M	W/m ²
	Radiance	L	W/m ² sr

Capacity for radiation within a specified spectral band to do work

Measuring energy

Term	Symbol	Units	
Radiant energy	Q	J (Joules)	
Radiant flux	ϕ	W (Watts, J/s)	
Radiant flux density	Irradiance	E	W/m ²
	Radiant exitance	M	W/m ²
Radiance	L	W/m ² sr	

Time rate of energy onto, off of, or through a surface

Measuring energy

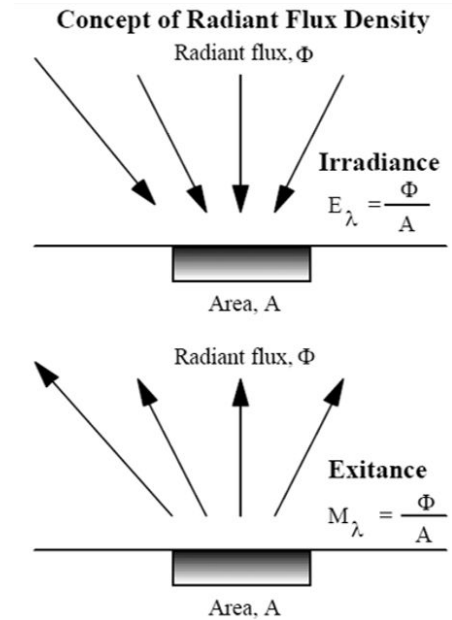
Term	Symbol	Units	
Radiant energy	Q	J (Joules)	
Radiant flux	ϕ	W (Watts, J/s)	
Radiant flux density	Irradiance	E	W/m ²
	Radiant exitance	M	W/m ²
Radiance	L	W/m ² sr	

Radiant flux upon a surface per unit area

Radiant flux leaving a surface per unit area

Measuring energy

	Term	Symbol	Units
	Radiant energy	Q	J (Joules)
	Radiant flux	ϕ	W (Watts, J/s)
Radiant flux density	Irradiance	E	W/m ²
	Radiant exitance	M	W/m ²
	Radiance	L	W/m ² sr



Measuring energy

	Term	Symbol	Units
	Radiant energy	Q	J (Joules)
	Radiant flux	ϕ	W (Watts, J/s)
Radiant flux density	Irradiance	E	W/m ²
	Radiant exitance	M	W/m ²
	Radiance	L	W/m ² sr

Remote sensing is the quantification and study of radiance.

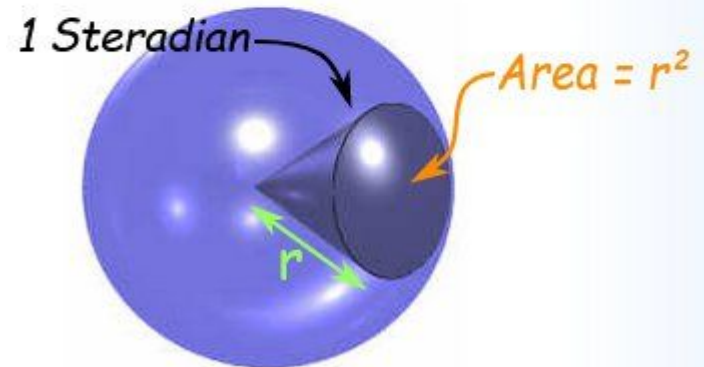
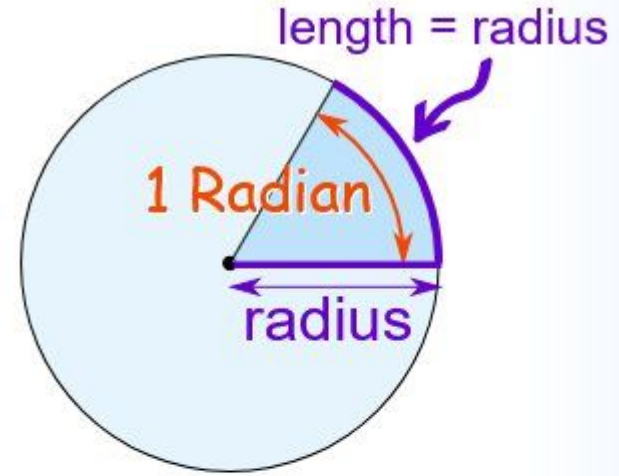
Measuring energy

	Term	Symbol	Units
	Radiant energy	Q	J (Joules)
	Radiant flux	ϕ	W (Watts, J/s)
Radiant flux density	Irradiance	E	W/m ²
	Radiant exitance	M	W/m ²
	Radiance	L	W/m ² sr

Remote sensing is the quantification and study of radiance.

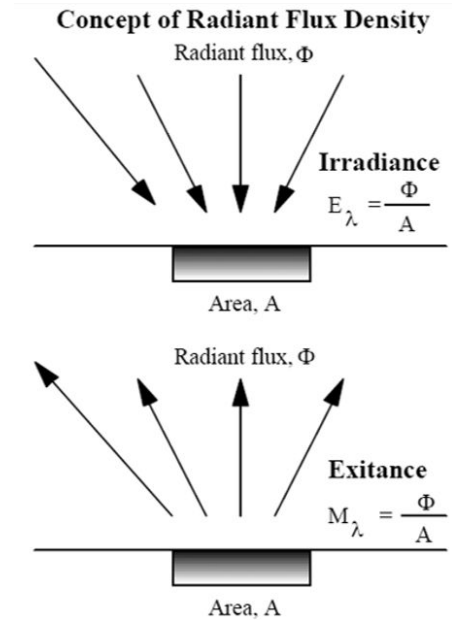
Measuring energy: steradians

- An angle in radians, projected onto a circle, gives a length
- A solid angle in steradians, projected onto a sphere, gives an area on the surface

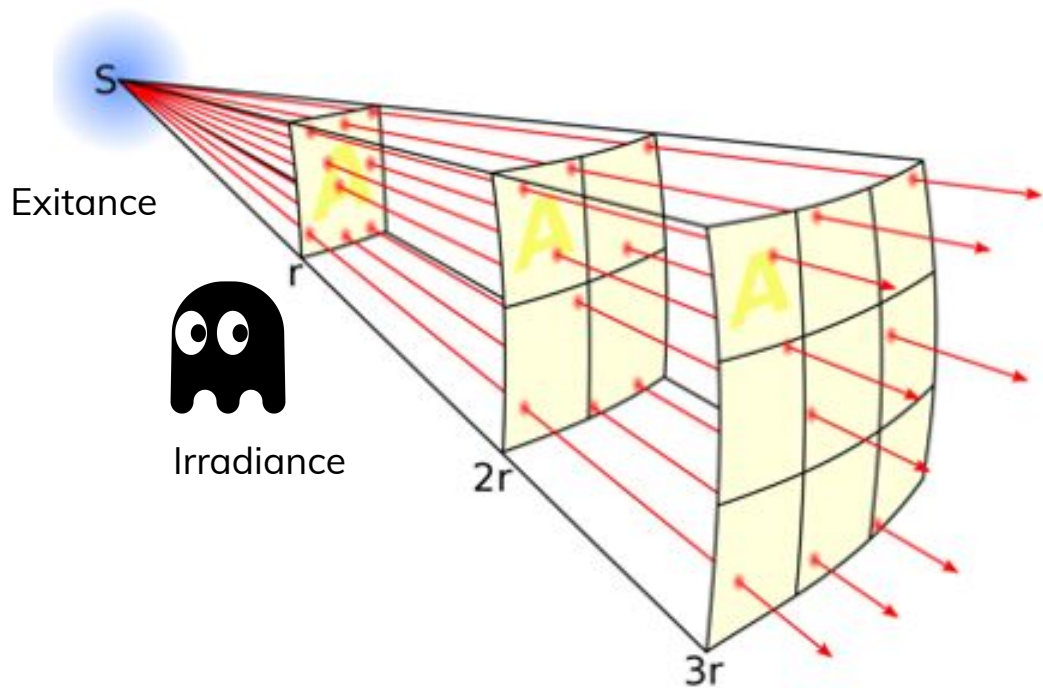


Measuring energy: exitance, irradiance, and radiance

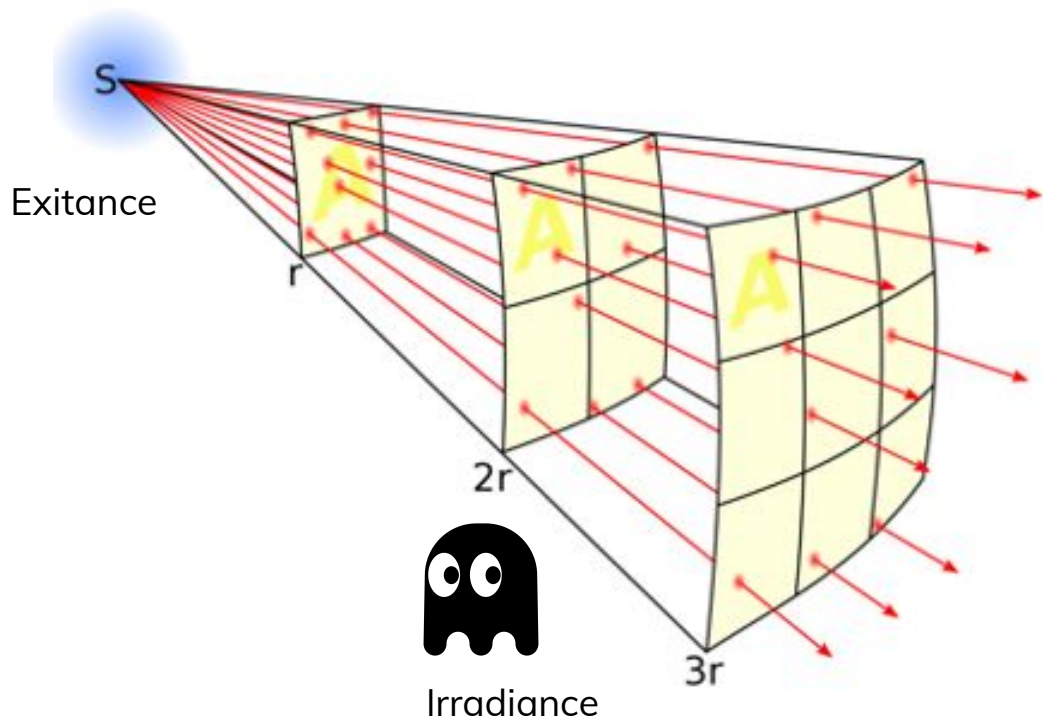
Term	Symbol	Units	
Radiant energy	Q	J (Joules)	
Radiant flux	ϕ	W (Watts, J/s)	
Radiant flux density	Irradiance	E	W/m^2
	Radiant exitance	M	W/m^2
Radiance	L	W/m^2sr	



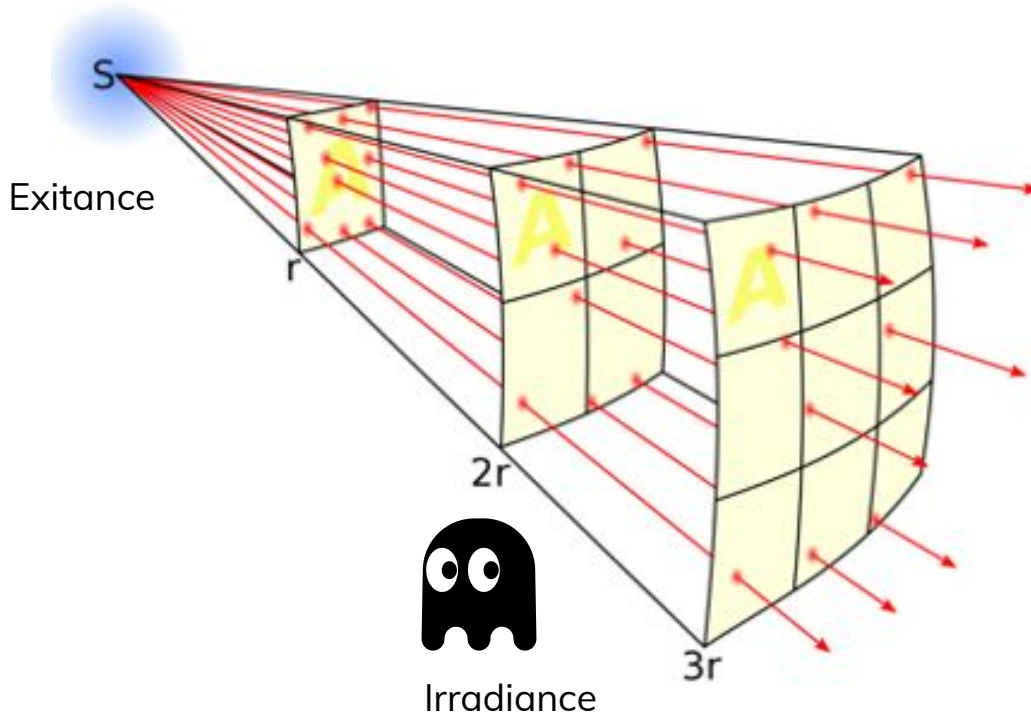
Measuring energy: irradiance vs. radiance



Measuring energy: irradiance vs. radiance

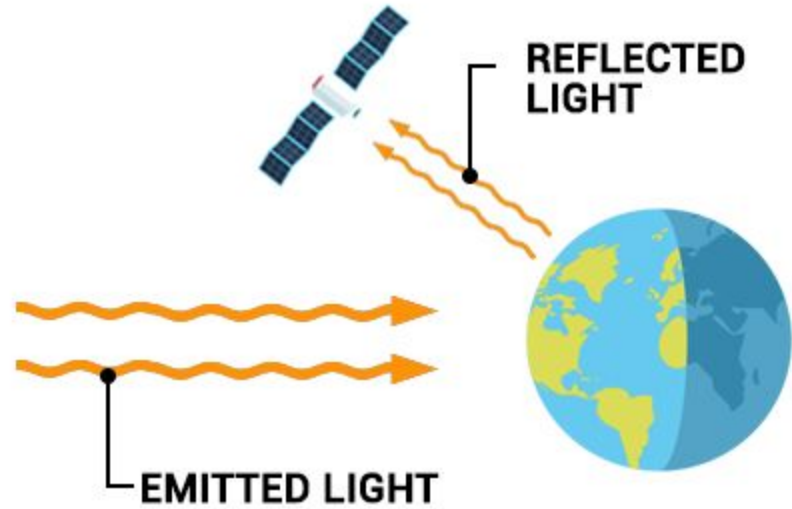


Measuring energy: irradiance vs. radiance

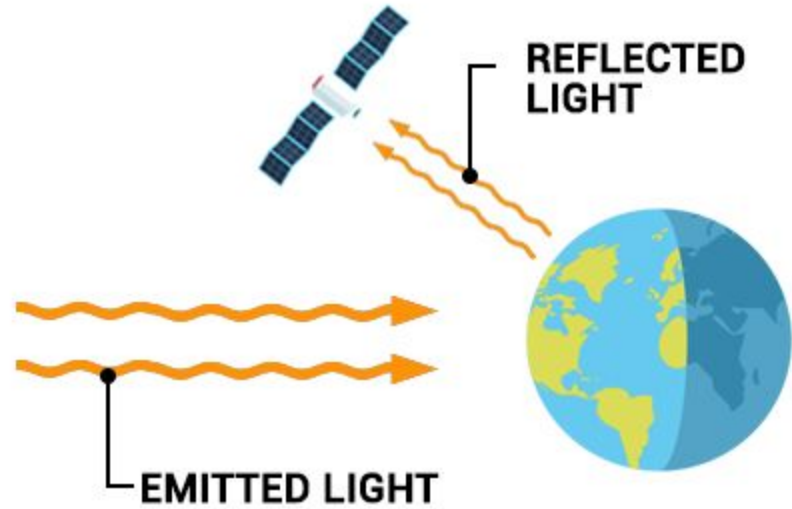
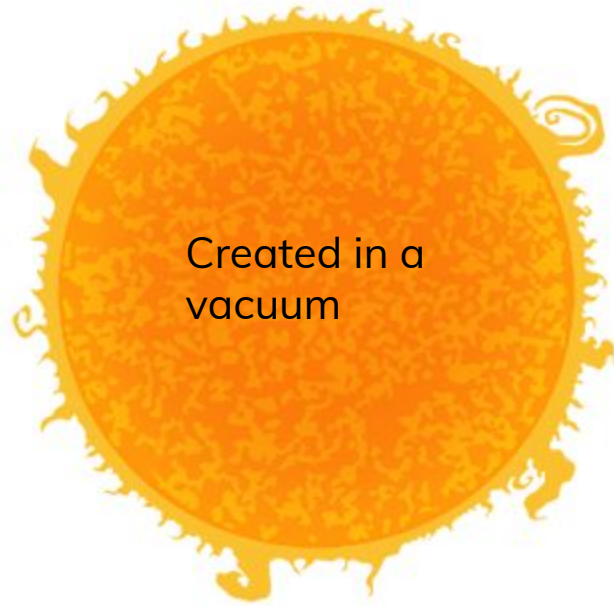


- Even when exitance stays constant, irradiance falls off with distance
- Normalizing by steradian means that radiance stays constant
 - Radiance does not change with distance
 - Makes remote sensing possible!

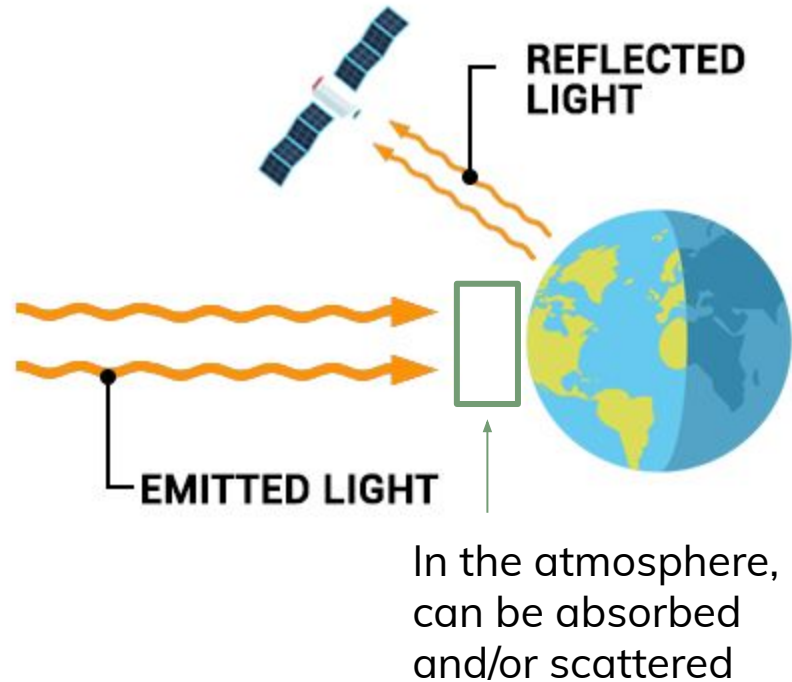
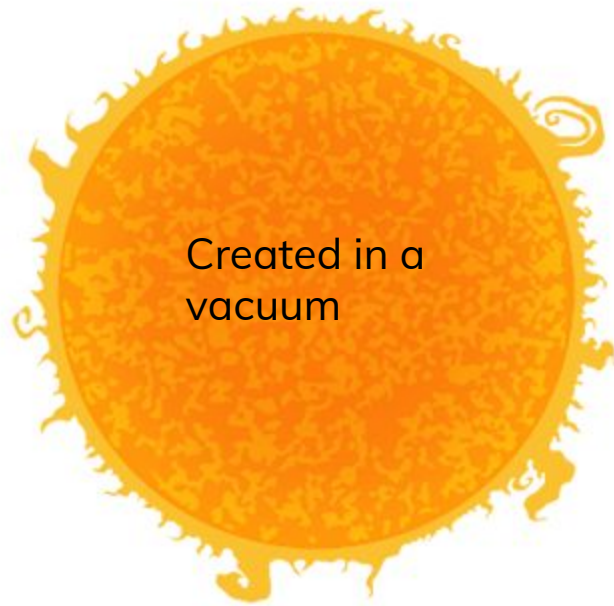
Radiation budget



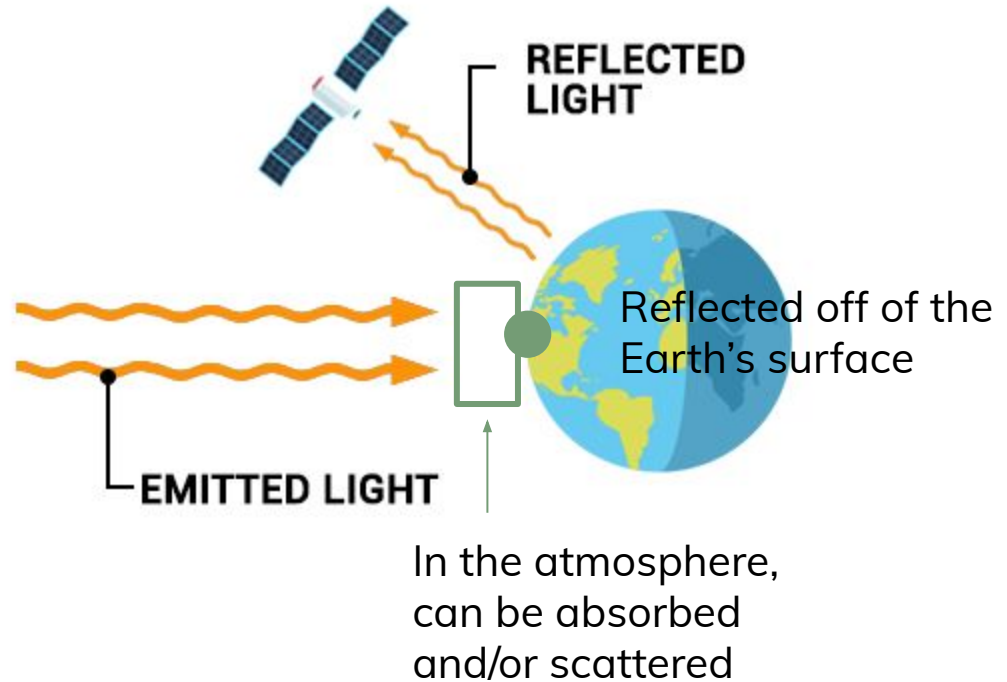
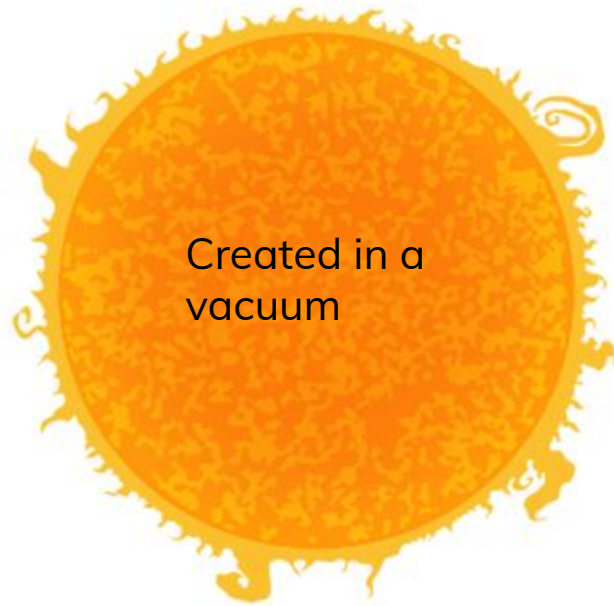
Radiation budget



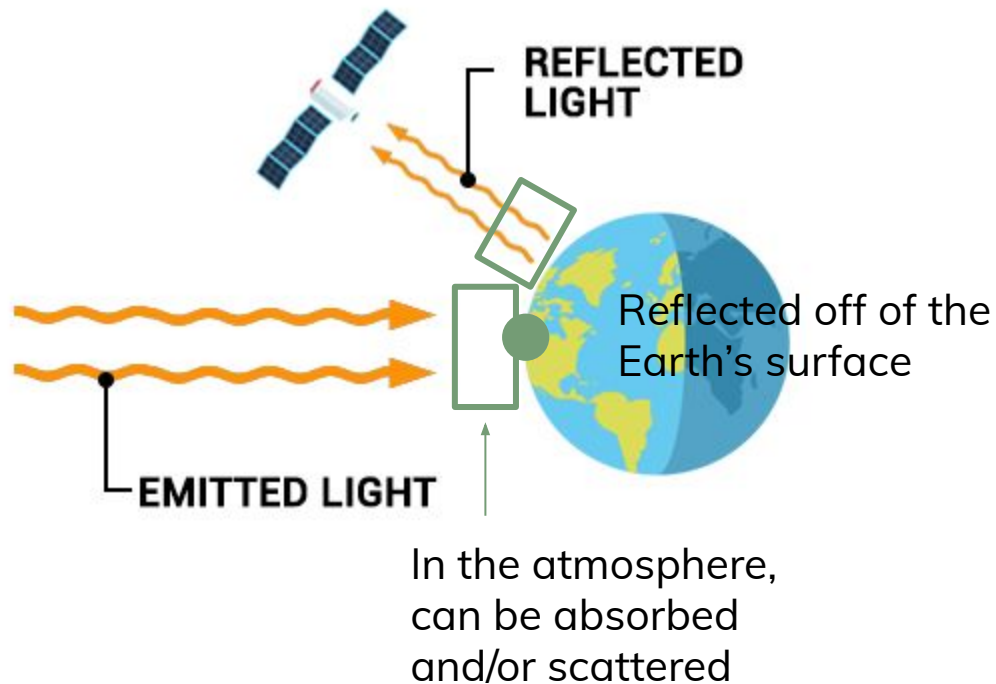
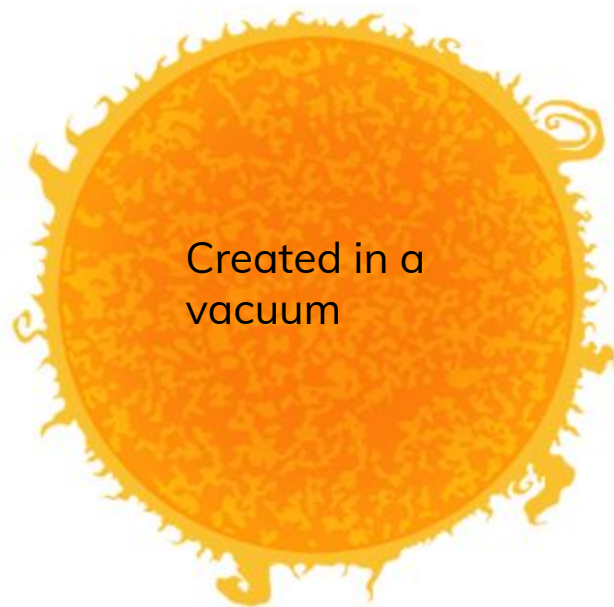
Radiation budget



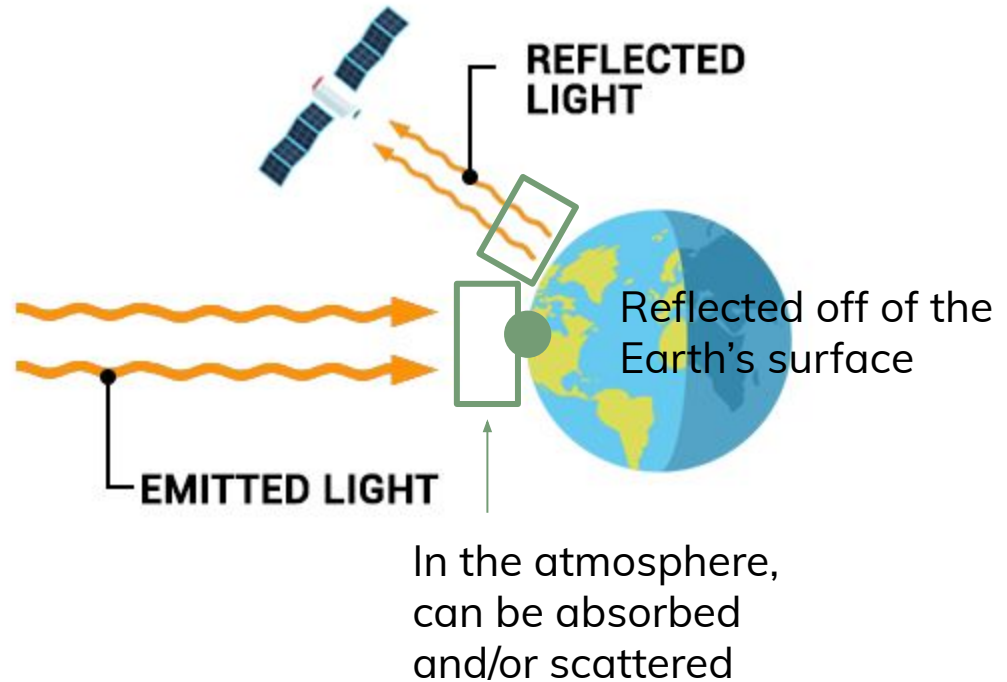
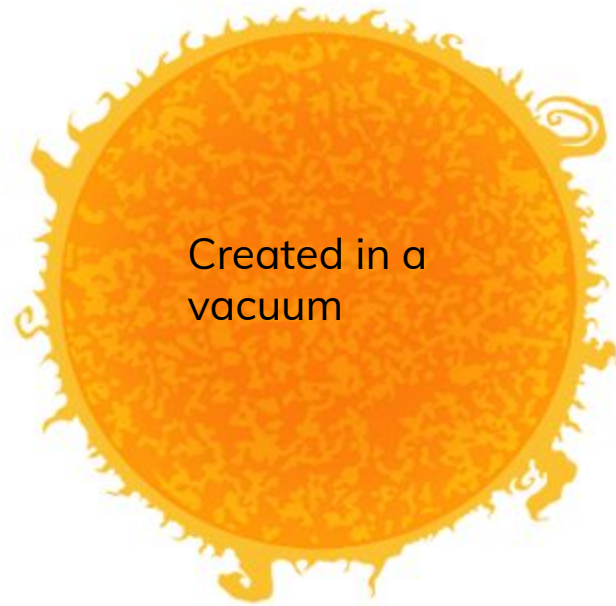
Radiation budget



Radiation budget

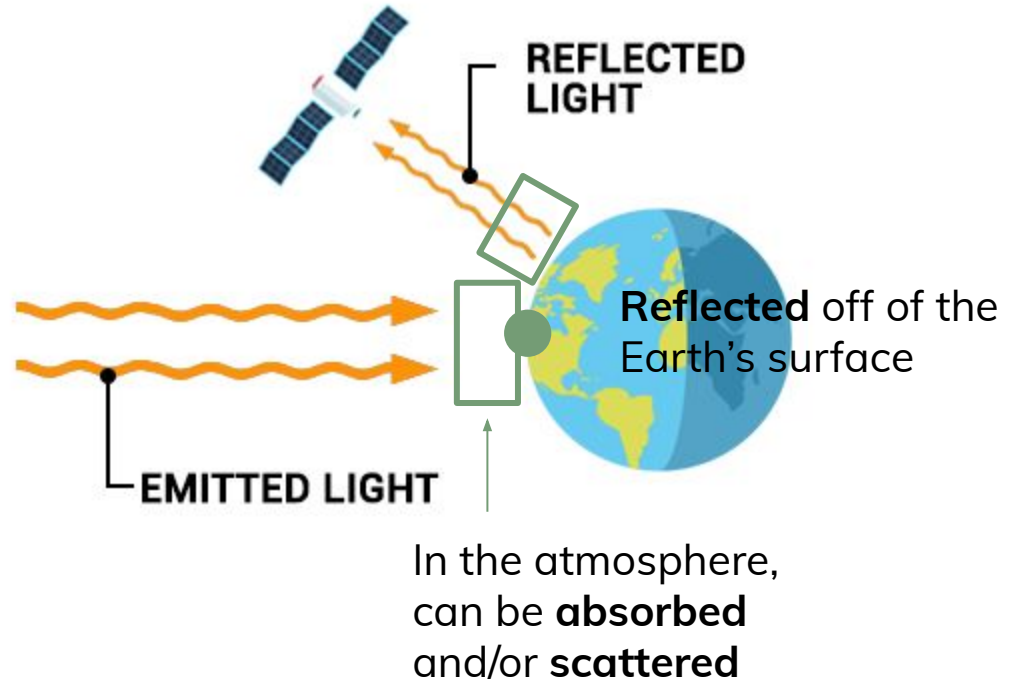
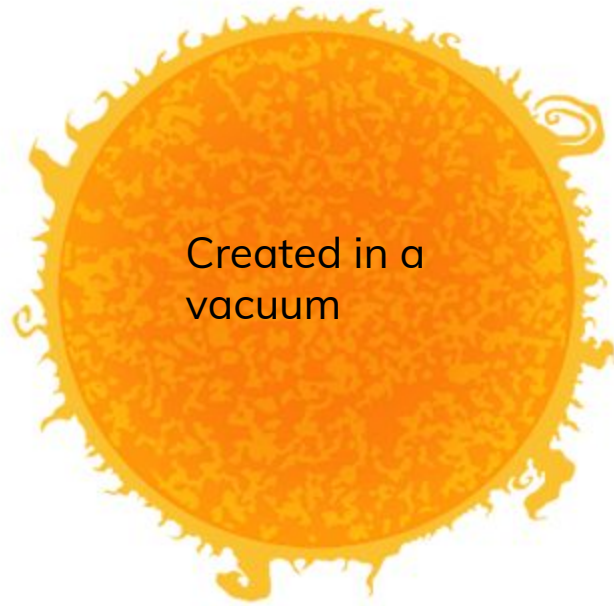


Radiation budget



By passing through media of different densities, can be refracted

Radiation budget



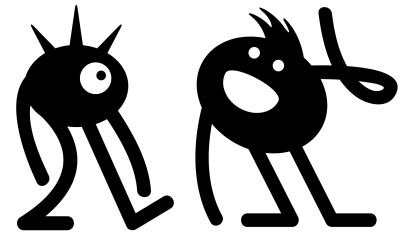
By passing through media of different densities, can be **refracted**

Absorption

- The process by which radiant energy is absorbed and converted into other forms of energy

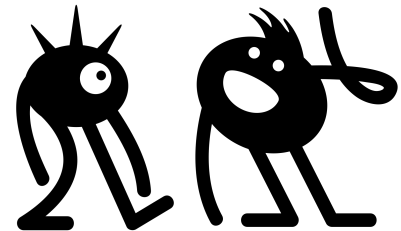
Absorption

- The process by which radiant energy is absorbed and converted into other forms of energy
- Name the top 3 atmospheric constituents which absorb radiation:

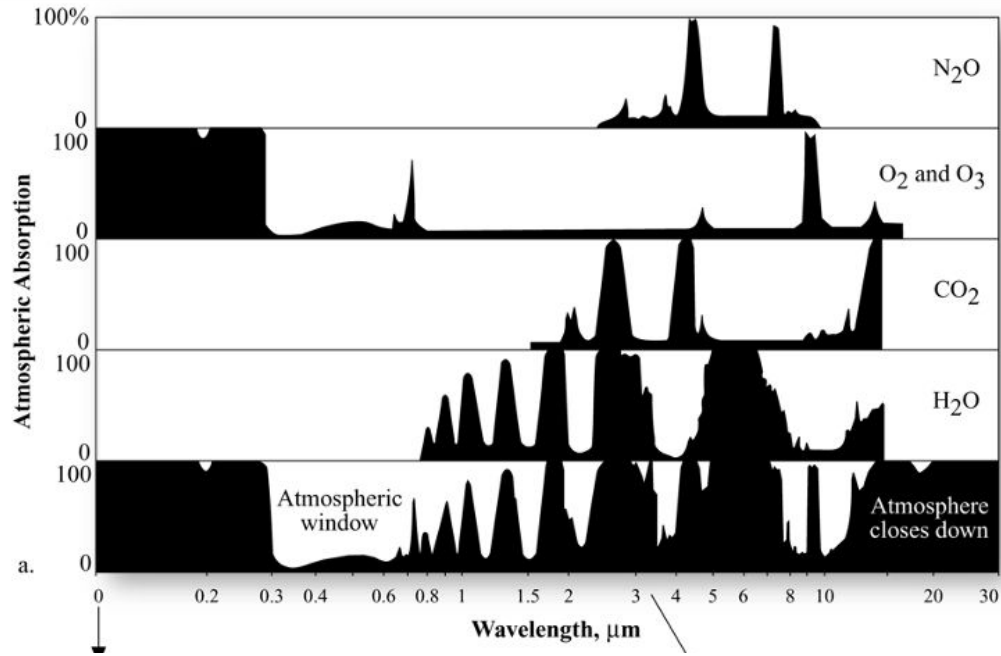


Absorption

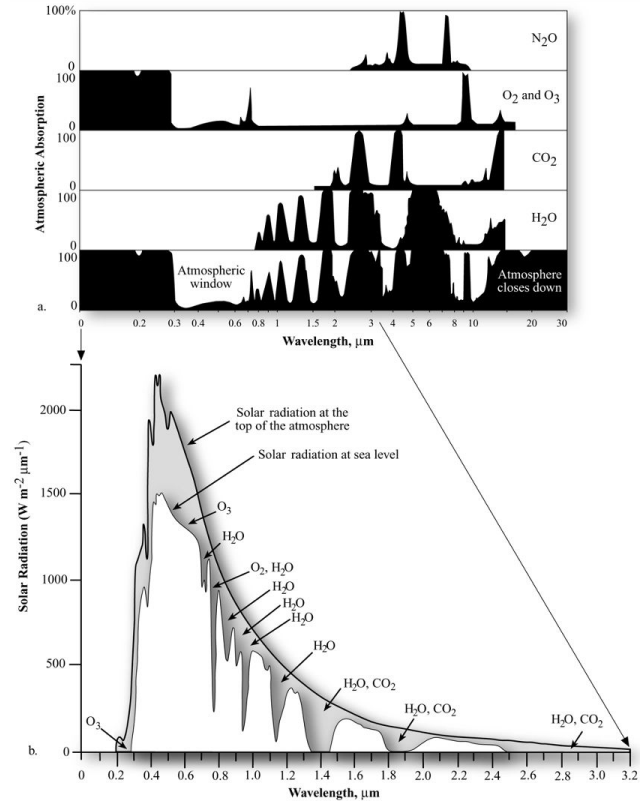
- The process by which radiant energy is absorbed and converted into other forms of energy
- Name the top 3 atmospheric constituents which absorb radiation:
 - Ozone
 - Carbon dioxide
 - Water vapor



Absorption



Absorption

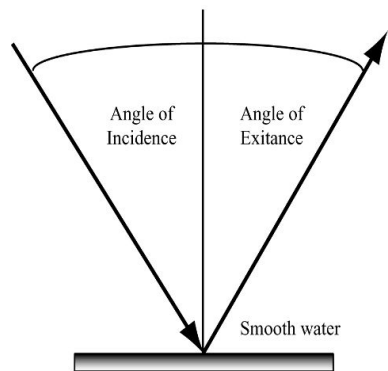


Reflectance

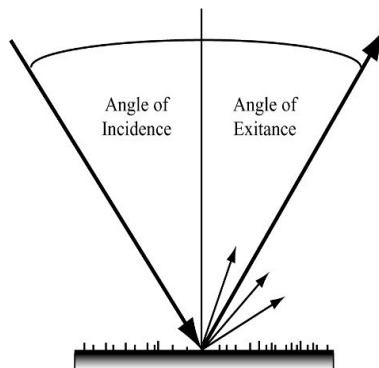
- The process whereby radiation “bounces off” an object and experiences no change in wavelength or frequency

Reflectance

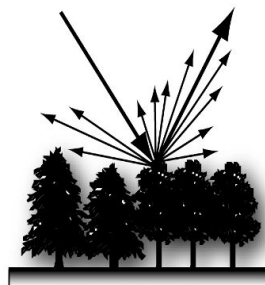
Specular versus Diffuse Reflectance



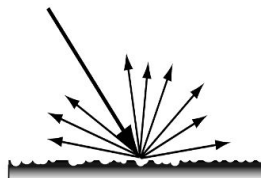
a. Perfect specular reflector.



b. Near-perfect specular reflector.

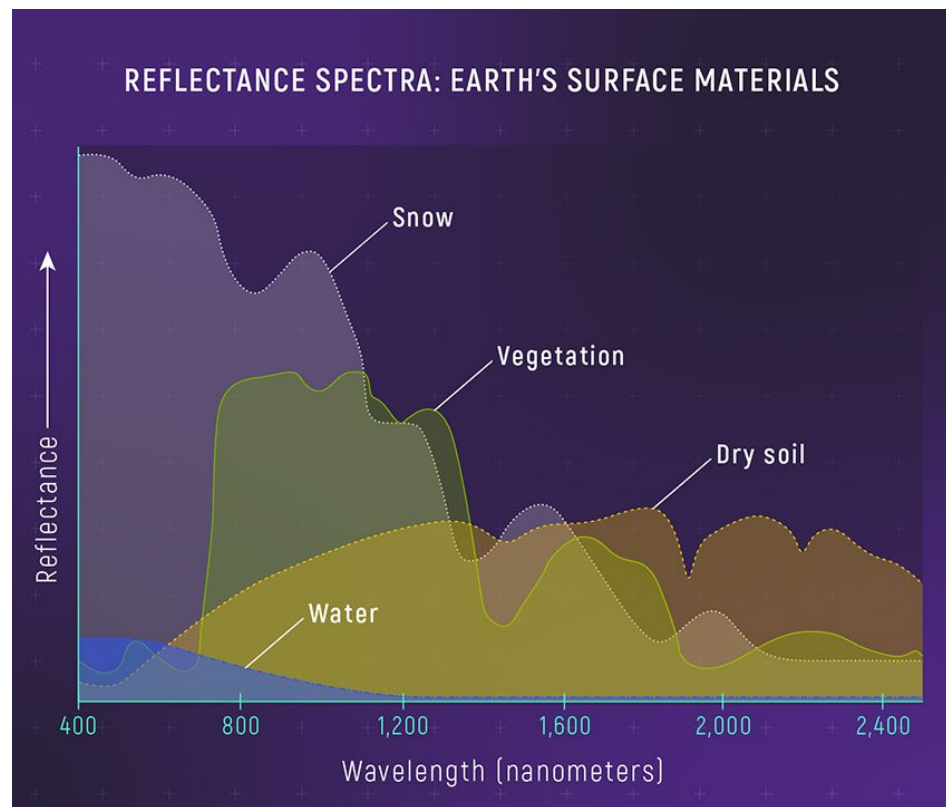


c. Near-perfect diffuse reflector.



d. Perfect diffuse reflector, or Lambertian surface.

Reflectance



Scattering

- Reflectance in an *unpredictable* manner
- Amount of scattering depends on:
 - Amount and size of particles or gases radiation is interacting with
 - Wavelength of radiation
 - Distance that radiant energy travels through atmosphere

Scattering

Three types of scattering:

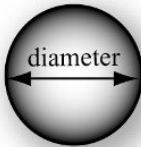
- **Rayleigh scattering**
 - Particle size $\ll \lambda_{\text{light}}$
 - Highly dependent on wavelength
- **Mie scattering**
 - Particle size $\sim \lambda_{\text{light}}$
 - Not strongly dependent on wavelength
- **Non-selective scattering**
 - Particle size $\gg \lambda_{\text{light}}$

Atmospheric Scattering


Rayleigh Scattering

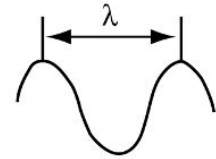
a.  Gas molecule

Mie Scattering

b.  Smoke, dust

Nonselective Scattering

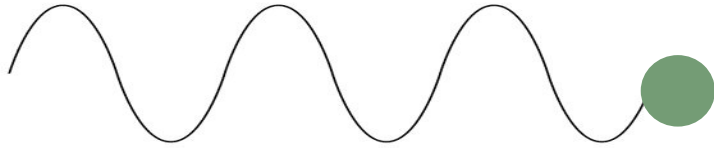
c.  Water vapor



Photon of electromagnetic energy modeled as a wave

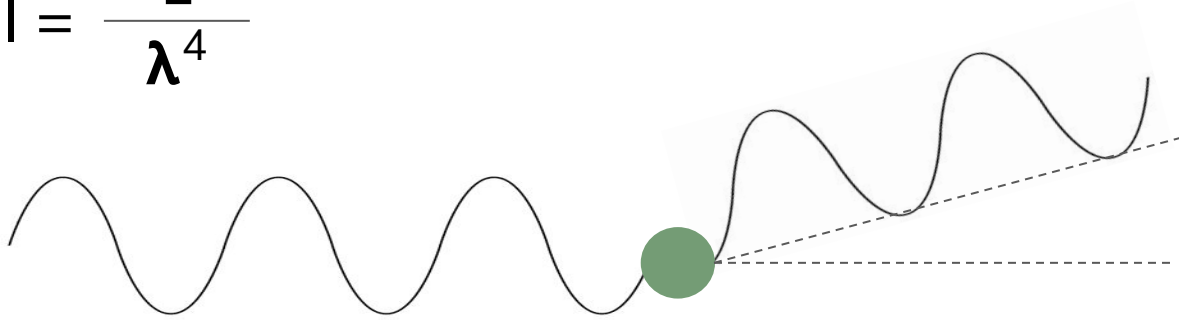
Rayleigh scattering

$$I = \frac{1}{\lambda^4}$$



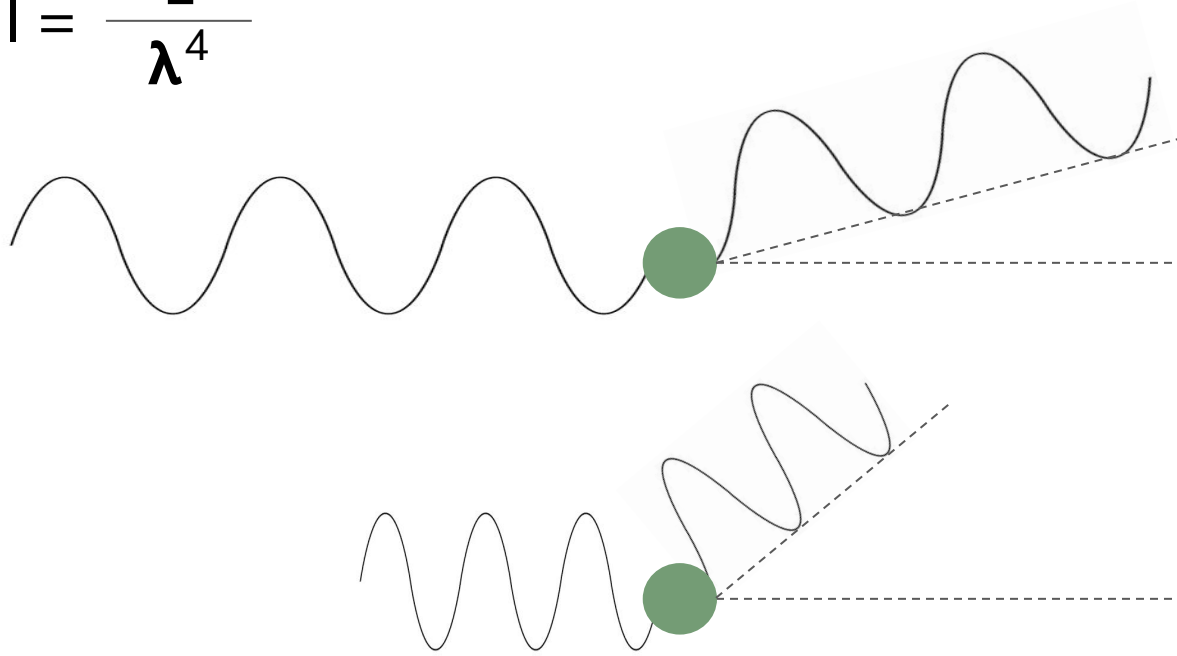
Rayleigh scattering

$$I = \frac{1}{\lambda^4}$$



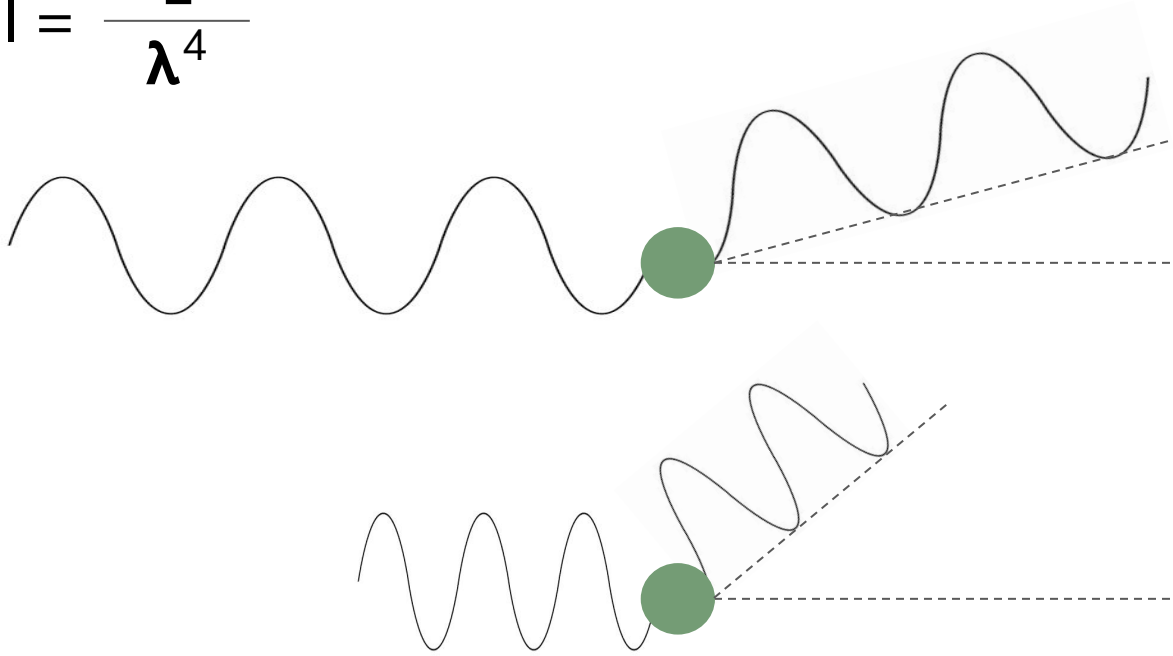
Rayleigh scattering

$$I = \frac{1}{\lambda^4}$$



Rayleigh scattering

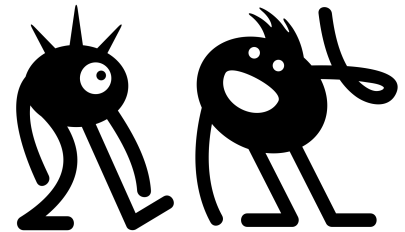
$$I = \frac{1}{\lambda^4}$$



As wavelength increases, intensity of scattering decreases

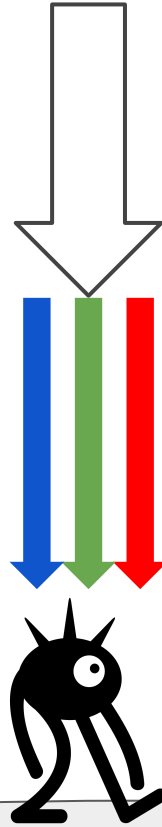
Rayleigh scattering

- Why is the sky blue?
- Why are sunsets red?



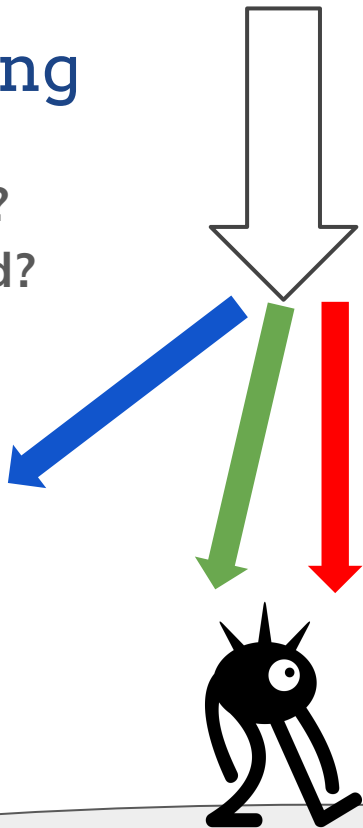
Rayleigh scattering

- Why is the sky blue?
- Why are sunsets red?



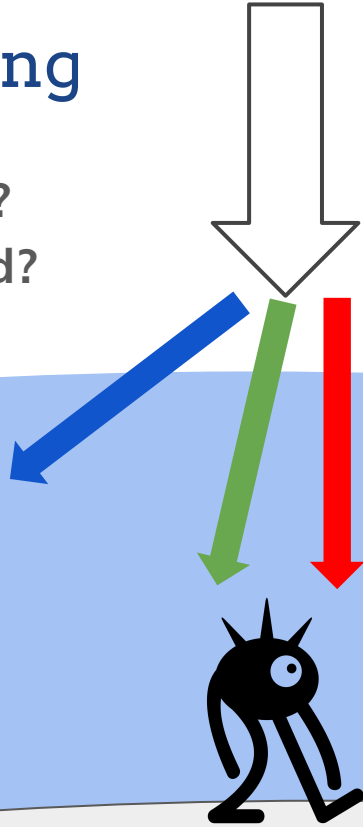
Rayleigh scattering

- Why is the sky blue?
- Why are sunsets red?



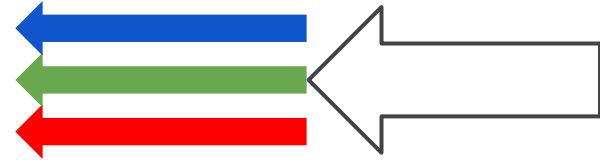
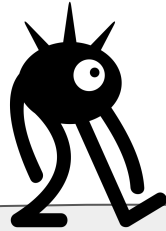
Rayleigh scattering

- Why is the sky blue?
- Why are sunsets red?



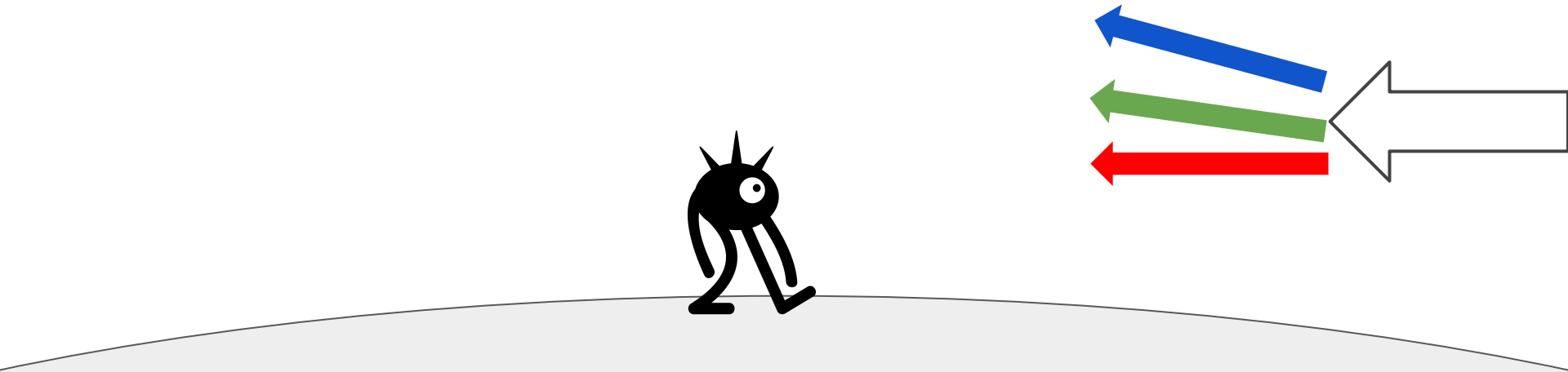
Rayleigh scattering

- Why is the sky blue?
- Why are sunsets red?



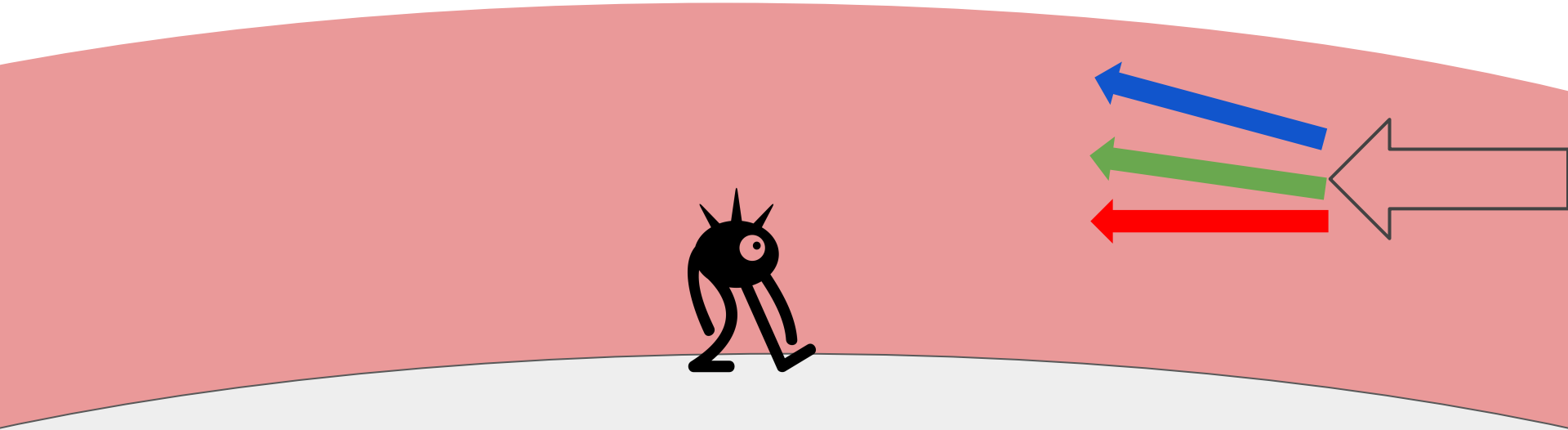
Rayleigh scattering

- Why is the sky blue?
- Why are sunsets red?



Rayleigh scattering

- Why is the sky blue?
- Why are sunsets red?



Mie and non-selective scattering

- **Mie scattering**
 - Amplifies wavelengths of similar size to particle
 - Pollution and aerosols scatter blue and green light away, contributing to red sunsets
- **Non-selective scattering**
 - Particles in the atmosphere several times the diameter of the wavelength
 - All wavelengths are scattered
 - Water droplets scatter all wavelengths of visible light equally well
 - Why clouds are white!

Refraction

- Refraction is ‘bending’ of light when it passes from one medium to another of different density.
 - The speed of EMR changes
 - In a vacuum $c \approx 3 \times 10^8$ m/s
- *Frequency of a light wave in a medium is determined by its source and is unaffected by the medium!*



Energy-matter interactions with terrain

Absorption: process by which radiation is absorbed and converted to other forms of energy.

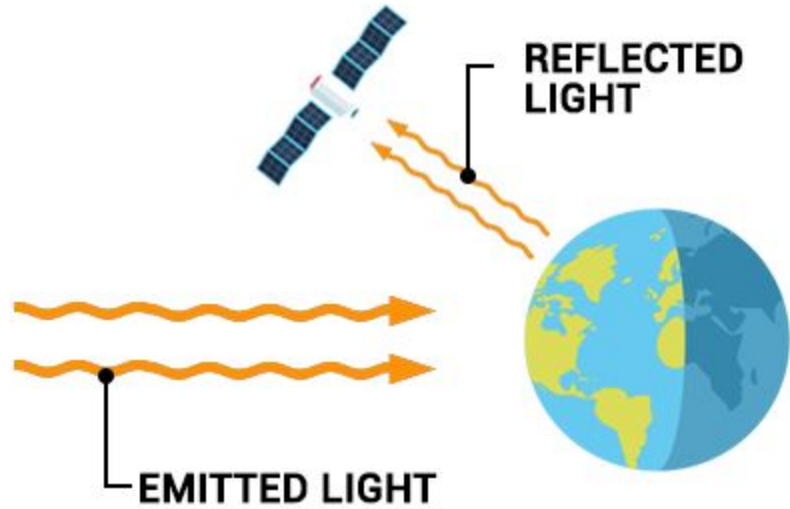
Reflectance: process whereby radiation “bounces off” an object.

Scattering: reflectance in an unpredictable manner.

Refraction: bending of light through mediums of different density.

Transmittance: process by which radiation passes through a material.

Create your own path diagram



Half way done!

R learners,

